CHM 2210 Lab: Melting Point

Introduction

The melting point of a solid is the temperature at which the solid melts (changes to a liquid) at 1 atm of pressure. Melting point is determined by slow heating of a small solid sample placed in a capillary tube. The melting point should be reported as a melting point range: from the temperature at which the first droplet of liquid is observed

A pure solid (pure enough for most purposes) usually has a sharp melting point which means that it melts within a range of 2°C or less.

An impure solid melts at a broader range (3°C or more) and at depressed temperature.

to the temperature at which the sample becomes a clear liquid, for example: 96.7 - 98.9°C.

Using melting point to find the identity of a solid is not sufficient because there are too many substances with the same melting points.

However, you can use a mixed melting point as a tool for identification of a solid.

If you mix two different solids with the same melting point, the melting point of the mixture will be broad and depressed. The melting point of a solid and a mixture of this solid with the authentic solid should be the same.

Example:

Measured/Observed melting point of an unknown solid X: 128.0 - 129.5°C

Melting point of known solid A: 130°C

Melting point of known solid B: 158°C

Measured melting point of mixture X + A: 127.5 - 129.0°C

Measured melting point of mixture X + B: 133.5 - 142.0°C

Identity of unknown X: solid A

In the future you will measure melting points of solids mostly to determine their purity.

Preparation of Sample

A solid sample should be dried and crushed with a metal spatula or bottom of a test tube.

If a solid is used for mixed melting point, the 1:1 mixture should be crushed together into a homogeneous mixture.

The pulverized solid should be mounded up and a small (1 - 2 mm) plug of sample should be pushed into the open end of the capillary tube. The sample should be knocked down to the sealed end of the capillary tube by tapping its scaled end on the surface or dropping it into a long glass tubing.

The sample should be 1-2 mm long and packed firmly.

If the sample is too large, the measured melting point may be too high.

If the sample is loosely packed, it may be difficult to determine the melting point range because of the air present.

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Gourmet Lab Sarah Reeves Young, 2011 Hands on inquiry based and relevant to every student OCOs life Gourmet Lab serves up a full menu of activities for science teachers of grades 60Co12 This collection of 15 hands on experiments0Coeach of which includes a full set of both student and teacher pagesOCochallenges students to take on the role of scientist and chef as they boil bake and toast their way to better understanding of science concepts from chemistry biology and physics By cooking edible items such as pancakes and butterscotch students have the opportunity to learn about physical changes in states of matter acids and bases biochemistry and molecular structure The Teacher pages include Standards addressed in each lab a vocabulary list safety protocols materials required procedures data analysis student questions answer key and conclusions and connections to spur wrap up class discussions Cross curricular notes are also included to highlight the lessonOCOs connection to subjects such as math and literacy Finally optional extensions for both middle school and high school levels detail how to explore each concept further What better topic than food to engage students to explore science in the natural world Fostering Sustained Learning Among Undergraduate Students: Emerging Research and Opportunities Chemosit, Caroline, Rugutt, John, Rugutt, Joseph K., 2017-03-03 Keeping students engaged and receptive to learning can at times be a challenge However by the implementation of new methods and pedagogies instructors can strengthen the drive to learn among their students Fostering Sustained Learning Among Undergraduate Students Emerging Research and Opportunities is an essential publication for the latest scholarly information on methods to inculcate student learning with a focus on implications to institutional policy and practices Featuring coverage on topics such as financial aid student motivation and mentorship this book is ideally designed for academicians practitioners and researchers seeking novel perspectives on the learning process and instruction methods New Coordinated Science: Chemistry Students' Book RoseMarie Gallagher, R. Gallagher, Paul Ingram, 2001-07-19 Provides information in manageable chunks which is reinforced by questions and activities that encourage students to consider the practical application of science to everyday life This work is useful for Higher Tier GCSE students Catalogue of the Ann Arbor High School for the Academic Year Ann Arbor High School (Mich.),1903 **Chemistry** Dorin, Prentice Hall Dictionary Editors, 1992 Michigan School Moderator, 1908

The Internet of Things for Education Linda Daniela, 2021-12-02 This book is about the Internet of Things in the field of education Specifically it focuses on two major topics IoT Internet of Things solutions to support distance education and new pedagogical approaches to support development of computational thinking with educational devices possessing the characteristics of IoT As the educational landscape has dramatically changed in times of global pandemic online resources and media such as IoT have become increasingly important This situation compels all educational scholars researchers and practitioners to search for new solutions new educational pathways and new agents for knowledge development to support learning This book presents the possibilities of IoT as both a catalyst and performance tool for education The convergence of

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