

## PHYSICAL KINETIC MOLECULAR THEORY

### Learning Objectives

- to identify theoretical differences in how gases are represented in the kinetic theory and in molecular models;
- to explain the relationship of the postulates of the kinetic molecular theory;
- to explain how molecular models affect fundamental theories;
- to represent quantitative information using graphical or tabular representation.

### Model 1: Representation of Gases in Different Phases

The following figure illustrates the representation of a diatomic gas in three phases: solid, liquid, and gaseous.



Figure 1: Representation of a diatomic gas in three phases: Solid, Liquid, and Gaseous state.

Model 1 illustrates the representation of a diatomic gas in three phases: Solid, Liquid, and Gaseous state. The diagram shows the particles in each phase and the differences in their arrangement and spacing.

### Key Questions

- What are the key differences in the representation of a diatomic gas in three phases: Solid, Liquid, and Gaseous state? How do these differences affect the properties of the gas?

Table 1: Representation of a diatomic gas in three phases: Solid, Liquid, and Gaseous state.

	SOLID	LIQUID	GASEOUS
<b>What are the key differences in the representation of a diatomic gas in three phases: Solid, Liquid, and Gaseous state?</b>	Particles are packed closely together in a regular arrangement. The particles are in contact with each other.	Particles are packed closely together in a disordered arrangement. The particles are in contact with each other.	Particles are widely spaced and in a disordered arrangement. The particles are not in contact with each other.
<b>How do these differences affect the properties of the gas?</b>	The gas has a low density and a low compressibility. The gas has a high boiling point and a high melting point.	The gas has a moderate density and a moderate compressibility. The gas has a moderate boiling point and a moderate melting point.	The gas has a high density and a high compressibility. The gas has a low boiling point and a low melting point.

# Lw Pogil Kinetic Molecular Theory Key

**L Towner**



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