

# SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

(i) The product obtained in the following reaction

P 239  
11.2



- (a) Pent-3-en-1-ol      ✓(b) Pentan-1-ol  
(c) Pentan-2-ol      (d) Pentanal

(ii) Amongst the following, the solubility of which ionic solid decreases with increase in temperature?

P 30  
11)

- (a)  $\text{KNO}_3$       (b)  $\text{NaBr}$   
✓(c)  $\text{Na}_2\text{SO}_4$       (d)  $\text{KCl}$

(iii) The correct IUPAC name of  $\text{Na}_3 [\text{AlF}_6]$  is

P 196  
Table  
9.3

- ✓(a) Sodium hexafluoroaluminate (III)  
(b) Sodium hexafluoroaluminate (II)  
(c) Sodium hexafluoroaluminium (III)  
(d) Sodium hexafluoroaluminium (II)

(iv) Which of the following acids has highest  $\text{pK}_a$  value?

P 277  
Table  
12.9

- (a) Mono chloroacetic acid      (b) Dichloroacetic acid  
(c) Trichloroacetic acid      ✓(d) Acetic acid

(v) Number of carbon atoms present in isoprene unit is \_\_\_\_.

P 322  
15.3

- (a) 6      ✓(b) 5  
(c) 4      (d) 3

(vi) The colourless transition metal ion amongst the following is

P 173

- (a)  $\text{Cu}^+$       3d<sup>10</sup>      (b)  $\text{Cu}^{++}$   
(c)  $\text{Ni}^{++}$       (d)  $\text{Co}^{++}$

# Hsc Exam Chemistry Answer Paper

**F Rizvi**



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