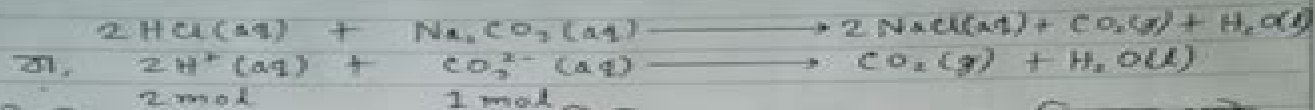


NAME OF THE EXPERIMENT: 0.1M Na_2CO_3	Page No.	
উদ্যম দ্বারা সংকলিত নমুনা উদ্যম (HCl/উদ্যম)	Expt No.	01
চলমান দিন	Date :	

মূলনীতি: Na_2CO_3 ও HCl এর প্রকৃত বিক্রিয়া সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃত বিক্রিয়া কোথায় প্রকৃত বিক্রিয়া পাবে, CO_3^{2-} এর মোল সংখ্যা (ফ্রাক্ট) $\equiv \text{H}^+$ এর মোল সংখ্যা (ক্যালি ব্রোমিডে দেয়া)

প্রয়োজনীয় সামগ্রিক তথ্য:

- ১) 0.1M Na_2CO_3 উদ্যম
- ২) 0.1M HCl উদ্যম
- ৩) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- ১) কনিক্যাল ফ্লাস্ক
- ২) ব্যালো
- ৩) পিপেট
- ৪) ক্যাম্পমই মেটার

কার্যের ধারা:

- ১) প্রস্তুত করা 0.1M HCl উদ্যম (প্রকৃত উদ্যম) অংশ নমুনা HCl দ্বারা ব্যালোটি বর্ণ করে এবং মেটারের সাথে ক্যাম্পমই মেটার দ্বারা পরিমাপ করা হবে।
- ২) প্রস্তুত করা 0.1M Na_2CO_3 উদ্যম (প্রকৃত উদ্যম) এর আনুমানিক ফ্রাক্টের সাথে পিপেট দ্বারা 10ml উদ্যম কনিক্যাল ফ্লাস্কে নিয়ে, কনিক্যাল ফ্লাস্কে উদ্যম 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করে, উদ্যম উদ্যমের বর্ণ হলুদ হয়।
- ৩) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা কাগজ টুকরা রাখি, ব্যালোটি HCl উদ্যম উদ্যমের দিকে নিয়ে ঘোঁরা দিতে 'গম পাঠ' - এ ফেরত করে, চিহ্নযুক্ত সাদা সাদা ব্যালোটিতে সাদা সাদা ব্যালোটিতে HCl উদ্যম ফোঁটা ফোঁটা কনিক্যাল ফ্লাস্কে যোগ করে

Hsc Chemistry 2nd Paper Practical 23

Marcel A. Müller



Hsc Chemistry 2nd Paper Practical 23:

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