Page 2	Mark Scheme	Syllabus	Paper
	IGCSE - May/June 2008	0620	31

An incorrectly written symbol, e.g. NA or CL, should be penalised once in a question.

- 1 (a) bromine [1]
  - (b) germanium [1]
  - (c) potassium or calcium [1]
  - (d) krypton [1]
  - (e) iron or cobalt [1]
  - (f) bromine [1]
  - (g) vanadium [1]

ACCEPT name or symbol

[Total: 7]

2 (a)

electron	e ore	1/1840 or 1/2000 or 0 1/1837 or negligible	-1
proton	p or p or H	1	*1
neutron	n	1	0 or neutral

- each correct row (1)
- (b) (i) equal numbers of protons and electrons of positive and negative charges or charges cancel/balance or net charge = 0
  [1]
  - (ii) lose electron(s)
    more protons than electrons
    NOT more + than -
  - (iii) different numbers of neutrons [1] same number of protons or same number of electrons [1] for just giving- they are isotopes [1] ONLY
  - (iv) an element is known for each proton number accept any sensible idea, for example no gaps between z = 1 and z = 103

[Total: 10]

# Marking Scheme Of 0620 32 M J 08

**Victor M. Corman** 

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