

(iii) Suggest why the mean bond enthalpy of a C=C bond is less than twice the mean bond enthalpy of a C—C bond.

(1)

(iv) Use the mean bond enthalpy data in the table, and the equation given below, to calculate a value for the standard enthalpy change of combustion of propene.

(3)

Bond	Mean bond enthalpy / kJ mol ⁻¹
C=C	612
C—C	347
C—H	413
O=O	498
C=O	805
O—H	464



Answer = _____ kJ mol⁻¹



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1. Understanding the eBook Jan 2015 Ial Core 12 Edexcel
 - The Rise of Digital Reading Jan 2015 Ial Core 12 Edexcel
 - Advantages of eBooks Over Traditional Books
2. Identifying Jan 2015 Ial Core 12 Edexcel
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jan 2015 Ial Core 12 Edexcel
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jan 2015 Ial Core 12 Edexcel
 - Personalized Recommendations
 - Jan 2015 Ial Core 12 Edexcel User Reviews and Ratings
 - Jan 2015 Ial Core 12 Edexcel and Bestseller Lists
5. Accessing Jan 2015 Ial Core 12 Edexcel Free and Paid eBooks

- Jan 2015 Ial Core 12 Edexcel Public Domain eBooks
- Jan 2015 Ial Core 12 Edexcel eBook Subscription Services
- Jan 2015 Ial Core 12 Edexcel Budget-Friendly Options
- 6. Navigating Jan 2015 Ial Core 12 Edexcel eBook Formats
 - ePub, PDF, MOBI, and More
 - Jan 2015 Ial Core 12 Edexcel Compatibility with Devices
 - Jan 2015 Ial Core 12 Edexcel Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jan 2015 Ial Core 12 Edexcel
 - Highlighting and Note-Taking Jan 2015 Ial Core 12 Edexcel
 - Interactive Elements Jan 2015 Ial Core 12 Edexcel
- 8. Staying Engaged with Jan 2015 Ial Core 12 Edexcel
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jan 2015 Ial Core 12 Edexcel
- 9. Balancing eBooks and Physical Books Jan 2015 Ial Core 12 Edexcel
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jan 2015 Ial Core 12 Edexcel
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Jan 2015 Ial Core 12 Edexcel
 - Setting Reading Goals Jan 2015 Ial Core 12 Edexcel
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jan 2015 Ial Core 12 Edexcel
 - Fact-Checking eBook Content of Jan 2015 Ial Core 12 Edexcel
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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