

Chapter 22

Hydrocarbon Compounds

22.1 Hydrocarbons

Organic Chemistry and Hydrocarbons

- Some thought that only living organisms could synthesize the carbon compounds found in living cells.
- Synthetic chemical products (fossil fuels, plastics, dyes, etc.) and hydrocarbon derivatives are synthesized, processed, and commercialized in nature.
- Today, organic chemistry involves the discovery of almost all carbon compounds, regardless of origin.

Introduction to Hydrocarbons

Hydrocarbon compounds contain only hydrogen and carbon.

Alkanes are the simplest hydrocarbon compounds.

Alkanes contain the most carbon chains, as carbon atoms always form four covalent bonds.

Alkanes are the simplest hydrocarbon compounds and contain carbon atoms joined by single bonds only.

Alkanes are the simplest and most inert compounds of organic compounds. They contain the most carbon atoms in their molecules and are the most abundant.

They are the simplest and most inert compounds of organic compounds. They contain the most carbon atoms in their molecules and are the most abundant.



The Properties of Hydrocarbons

Hydrocarbons are the simplest compounds.

Hydrocarbons are the simplest compounds that contain carbon atoms. They are the simplest compounds that contain carbon atoms and are the most abundant.

Alkanes and Alkenes

Alkanes are the simplest and most inert compounds of organic compounds. They contain the most carbon atoms in their molecules and are the most abundant.

Hydrocarbon Compounds Ch 22 Study Guide

Gaurav Sinha



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