

Chapter 22

Hydrocarbons: Compounds

22.1 Hydrocarbons

Organic Chemistry and Hydrocarbons

- Know through what only being organic is distinguished: the carbon-hydrogen bond in their cells
- Natural chemical products are often made by very simple hydrocarbon molecules and structures, even in organic compounds of plants
- Today, organic chemistry includes the synthesis of almost all carbon compounds, regardless of origin

Introduction to Hydrocarbons

Hydrocarbon compounds are composed of hydrogen and carbon

Alkanes is an important class of hydrocarbons

Alkanes contain the most carbon chains, as carbon atoms always form four covalent bonds

Alkanes are saturated hydrocarbons, containing only single bonds between all carbon atoms



Alkanes are the simplest of the most important of organic compounds. Alkanes make up a broad spectrum of natural and synthetic molecules.

They are the only hydrocarbons that have a single carbon-carbon bond. In this respect they are the most saturated hydrocarbons.

Alkanes and Hydrocarbons

Alkanes are saturated hydrocarbons

Alkanes are hydrocarbons that contain only single bonds between carbon atoms. They are the most saturated hydrocarbons and are the most important class of hydrocarbons.

Alkanes are the simplest of the most important of organic compounds.

Alkanes are the most saturated hydrocarbons, containing only single bonds between all carbon atoms.

Hydrocarbon Compounds Ch 22 Study Guide

Gaurav Sinha



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