



Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

Assessment Statements

- | | |
|--|--|
| <ul style="list-style-type: none"> ☐ Outline the Scientific Method ☐ Identify independent, dependent & controlled variables. ☐ Demonstrate correct and safe use of laboratory equipment. ☐ Distinguish between quantitative and qualitative data <ul style="list-style-type: none"> ➢ <i>Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.</i> ➢ <i>Quantitative is mathematical, including</i> <ul style="list-style-type: none"> ◊ <i>Raw data (as recorded)</i> ◊ <i>Processed (after calculations)</i> ☐ Distinguish between accuracy and precision <ul style="list-style-type: none"> ➢ <i>Accuracy describes the correctness of a set of measurements</i> ➢ <i>Precision describes the repeatability of a set of measurements</i> | <ul style="list-style-type: none"> ☐ State SI units of measurement for <ul style="list-style-type: none"> ➢ <i>Mass: grams (g)</i> ➢ <i>Length: Metres (m), millimeters (mm)</i> ➢ <i>Concentration: molarity (M)</i> ➢ <i>Temperature: Degrees Celsius (°C)</i> ➢ <i>Volume: Litres (L) or millilitres (mL)</i> ➢ <i>Density: grams per cubic centimeter (g/cm³)</i> ➢ <i>Conductivity: Siemens (S)</i> ➢ <i>Acidity or alkalinity (pH)</i> ☐ Energy: Joules (J or kJ)Determine the uncertainty of digital and analog measuring tools <ul style="list-style-type: none"> ➢ <i>Usually ± 1 of the smallest division</i> ☐ Calculate the % error of a given or recorded set of values <ul style="list-style-type: none"> ➢ <i>$(\text{Measured} - \text{True}) / \text{True} \times 100$</i> ☐ State values to appropriate numbers of significant figures ☐ Calculate values to appropriate numbers of significant figures |
|--|--|

Related Concepts:

- ☐ 01.1 Element symbols (Quia quiz: <http://www.quia.com/quiz/3246640.htm#>)
- ☐ 01.2 Accuracy, Precision, Error, Uncertainty (Quia quiz: <http://www.quia.com/quiz/3507626.htm#>)
- ☐ 01.3 SigDigs and Notation (Quia quiz: <http://www.quia.com/quiz/3504378.htm#>)

Command Terms: Be sure you understand what is being asked of you

Analyse Break down to identify key parts. Interpret information to reach conclusions.

Apply	Use knowledge & understanding in response to a real situation or given circumstances.
Calculate	Obtain a numerical answer, showing relevant stages in the working.
Define	Give the precise meaning of a word, phrase, concept or physical quantity.
Describe	Give a detailed account or picture of an event, pattern, process or pattern.
Design	Produce a plan, simulation or model.
Determine	Obtain the only possible answer.
Discuss	Offer a considered and balanced review that includes a range of arguments, factors or hypotheses. Opinions or conclusions should be presented clearly and supported by appropriate evidence.
Distinguish	Make clear the differences between two or more concepts or items.
Evaluate	Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria.
Explain	Give a detailed account including reasons or causes.
Identify	Provide an answers from a number of possibilities. Recognise and state briefly a distinguishing fact or feature.
Label	Add title, labels or brief explanation(s) to a diagram or graph.
List	Give a sequence of answers with no explanation.
Outline	Give a brief account.
Recall	Remember or recognize from prior learning experiences.
State	Give a specific name, value or other brief answer without explanation or calculation.
Suggest/Propose	Propose a solution, hypothesis or other possible answer.

Epcc Chemistry Lab Manual

Daniela Niemeyer



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mathematics engineering and technology Descriptions are organized by discipline and include projects in astronomy biology
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Dreizler, J.H. Macek, H.R.J. Walters, 1999-09-30 The last few years have seen some remarkable advances in the understanding
of atomic phenomena It is now possible to isolate atomic systems in traps measure in coincidence the fragments of collision
processes routinely produce and study multicharged ions One can look at bulk matter in such a way that the fundamental
atomic character is clearly evident and work has begun to tease out the properties of anti matter The papers in this book
reflect many aspects of modern Atomic Physics They correspond to the invited talks at a conference dedicated to the study of
New Directions in Atomic Physics which took place in Magdalene College Cambridge in July of 1998 The meeting was
designed as a way of taking stock of what has been achieved and it was hoped as a means of stimulating new research in new
areas along new lines Consequently an effort was made to touch on as many directions as we could in the four days of the
meeting We included some talks which overviewed whole subfields as well as quite a large number of research contributions
There is a unity to Physics and we tried to avoid any artificial division between theory and experiment We had roughly the
same number of talks from those who are primarily concerned with making measurements and from those who spend their
lives trying to develop the theory to describe the experiments **Acp Chem 1171 General Chemistry Lab Manual**
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