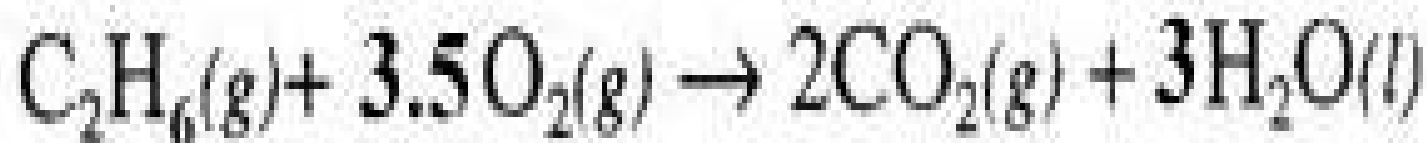


Calculate the enthalpy of combustion of ethane,  $\text{C}_2\text{H}_6$ , at  $25^\circ\text{C}$  and one atmosphere pressure.



| Compound                  | $\Delta H_f^\circ$ , $\text{kJ}\cdot\text{mol}^{-1}$ |
|---------------------------|------------------------------------------------------|
| $\text{C}_2\text{H}_6(g)$ | -84.7                                                |
| $\text{H}_2\text{O}(l)$   | -285.8                                               |
| $\text{CO}_2(g)$          | -393.5                                               |

# Entropy For Combustion Of Ethane C<sub>2</sub>H<sub>6</sub>

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