

Unit 3 Study-Guide: Test Study

1. Define the following terms as a separate sheet of paper:

Atom
Molecular Structure
Atomic Mass

Chemical Formula
Molecular Composition
Molarity

Volume Percent
M.M.
Molar Concentration

2. Find the mass ratio of the following:

H_2O
 H_2SO_4
 NaOH

3. Calculate the degree of acidity for the

4. Calculate the volume of H_2O present with H_2SO_4

5. Calculate the mass of H_2O present in 100 g of 10% solution of H_2SO_4

6. How many molecules are present in 100 g of H_2O

7. How many grams constitute 100 g of 10% solution of H_2SO_4

8. How many liters constitute 100 g of 10% solution of H_2SO_4

9. Find a solution containing 100 g of water. Calculate the molarity

10. Calculate the volume and mass of 100 g of 10% solution

11. Calculate the mass of H_2O in 100 g of 10% solution of H_2SO_4

Mass And The Mole Study Guide Answers

Jianjun Gao



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