



GoldStar

SERVICE MANUAL

NF-01X CHASSIS
CMT-4282

CAUTION

Before servicing the chassis, read the "Safety Precaution", "X-RAY Radiation Precaution" and "Product Safety Notice" in this manual.

SPECIFICATIONS

POWER INPUT 84 watts, AC 120 volts 60 Hz
ANTENNA INPUT IMPEDANCES: 300 ohm balanced (UHF)
75/300 ohm balanced (VHF)

RECEIVING CHANNELS
12 VHF channels 2 to 13
70 UHF channels 14 to 83
57 CATV channels 14-64, 95-99, 01

INTERMEDIATE FREQUENCIES
Picture I-F carrier frequency 45.75 MHz
Sound I-F carrier frequency 41.25 MHz
Color sub-carrier frequency 42.17 MHz
Center frequency 44 MHz

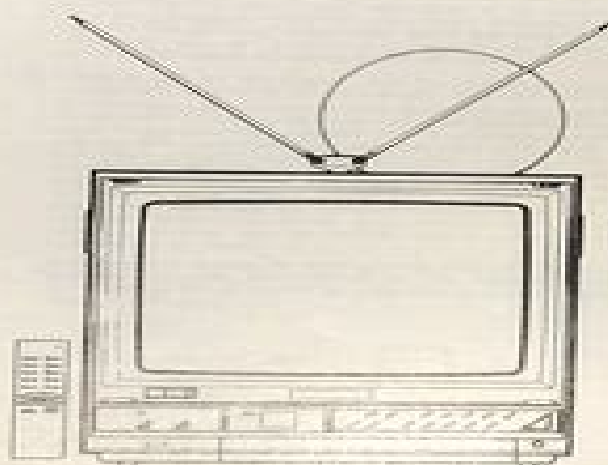
CHASSIS CONSTRUCTION IC-Solid state chassis
PICTURE TUBE Type No. 370FCB22
SOUND OUTPUT 0.8 Watt min. (at 10% harmonics
distortion) 1.0 watt max.

CABINET Plastic, portable

WEIGHT 14.1 Kg

DIMENSION Height 374 mm
Width 354 mm
Depth 365 mm

ANTENNAS VHF telescopic and UHF loop antenna
(Accessories)



CMT-4282

JUNE, 1985

P/N 483-211A
Printed in Korea

Goldstar Tv User Guide

Carol Costa, Ravel Buckley



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word exothermic is derived from exo the greek word for outside and therme the greek word

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web 1 sodium hydroxide solution is poured into a beaker of hydrochloric acid which contains a thermometer showing room temperature 2 the beaker now contains sodium chloride and water and the

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm³ 3 place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with hcl as 57.9 kJ mol⁻¹

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o l}$ for which $\Delta H = 57\text{ kJ mol}^{-1}$ eg 2 $\text{h}_2\text{so}_4 \text{ aq} + \text{koh aq} \rightarrow \text{ks}_2\text{so}_4 \text{ aq} + \text{h}_2\text{o l}$ for which $\Delta H = 57\text{ kJ mol}^{-1}$

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