Name: Amanda Brito Date: 04/29/2019

#### Student Exploration: Equilibrium and Concentration

Vocabulary: chemical equilibrium, concentration, equilibrium, equilibrium constant, reaction quotient, reversible reaction

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
Gary has \$5,000 in his bank account and earns a modest salary. Every month he pays for rent, food, utilities, and entertainment.

- A. How will Gary's account change if he saves more than he spends? \_it will increase.
- B. How will Gary's account change if he spends more than he saves? It will decrease.
- C. What happens if Gary spends exactly as much as he saves? It will stay the same.

#### Gizmo Warm-up

If Gary spends exactly as much as he earns, his savings will be in **equilibrium**. Equilibrium occurs when two opposing processes occur at the same rate, leading to no net change. In the Equilibrium and Concentration Gizmo™, you will investigate how equilibrium can occur in chemical reactions.

To begin, check that Reaction 1 is selected. Set Moles  $NO_2$  to 8 and Moles  $N_2O_4$  to 0.



Click Play ( ) and observe the colliding molecules. What do you notice?
 the molecules move and collide. Occasionally there is a blue or red flash.

In the Gizmo, a blue flash appears every time two reactants combine to form a product. A red flash appears every time a product dissociates into reactants.

Click Reset (□), and set Moles NO₂ to 0 and Moles N₂O₄ to 8. Click Play.

What do you notice now? There are more red flashes than blue.

When a reaction can proceed in either direction, it is a reversible reaction. Based on what
you have observed, is the synthesis of NO<sub>2</sub> into N<sub>2</sub>O<sub>4</sub> a reversible reaction? Explain.

I know the synthesis of NO2 into N2O4 is a reversible reaction because both the forward reaction (blue flashes) and the reverse reaction (red flashes) are observed.



# **Equilibrium And Concentration Gizmo Answers**

**L** Cohen

#### **Equilibrium And Concentration Gizmo Answers:**

**An Introduction to Equilibrium Problems** ,2016 You use your basic understanding of equilibrium systems to try to solve some problems You tackle two kinds of equilibrium problems ones in which you are asked to calculate the equilibrium constant for an equation and ones in which you are asked to find the equilibrium concentration of a reactant or product

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#### **Table of Contents Equilibrium And Concentration Gizmo Answers**

- 1. Understanding the eBook Equilibrium And Concentration Gizmo Answers
  - The Rise of Digital Reading Equilibrium And Concentration Gizmo Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Equilibrium And Concentration Gizmo Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Equilibrium And Concentration Gizmo Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Equilibrium And Concentration Gizmo Answers
  - Personalized Recommendations
  - Equilibrium And Concentration Gizmo Answers User Reviews and Ratings
  - Equilibrium And Concentration Gizmo Answers and Bestseller Lists
- 5. Accessing Equilibrium And Concentration Gizmo Answers Free and Paid eBooks
  - Equilibrium And Concentration Gizmo Answers Public Domain eBooks
  - Equilibrium And Concentration Gizmo Answers eBook Subscription Services

- Equilibrium And Concentration Gizmo Answers Budget-Friendly Options
- 6. Navigating Equilibrium And Concentration Gizmo Answers eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Equilibrium And Concentration Gizmo Answers Compatibility with Devices
  - Equilibrium And Concentration Gizmo Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Equilibrium And Concentration Gizmo Answers
  - Highlighting and Note-Taking Equilibrium And Concentration Gizmo Answers
  - Interactive Elements Equilibrium And Concentration Gizmo Answers
- 8. Staying Engaged with Equilibrium And Concentration Gizmo Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Equilibrium And Concentration Gizmo Answers
- 9. Balancing eBooks and Physical Books Equilibrium And Concentration Gizmo Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Equilibrium And Concentration Gizmo Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Equilibrium And Concentration Gizmo Answers
  - Setting Reading Goals Equilibrium And Concentration Gizmo Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Equilibrium And Concentration Gizmo Answers
  - Fact-Checking eBook Content of Equilibrium And Concentration Gizmo Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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