

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At standard atmospheric conditions, all gases have approximately the same kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The kinetic molecular theory is based on the assumption that all particles have the same average kinetic energy.)
- At standard atmospheric conditions, all gases have the same average kinetic energy.
- At standard atmospheric conditions, all gases have the same average kinetic energy.

Model 1: Representation of Gases in Different Phases

The following figure illustrates the representation of gases in different phases.



Figure 1: Representation of Gases in Different Phases

At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.

Key Concepts

- At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The kinetic molecular theory is based on the assumption that all particles have the same average kinetic energy.)
- At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.

Table 1: Representation of Gases in Different Phases

	GAS	LIQUID	CRYSTALLINE SOLID
MOLECULES ARE MOVING RANDOMLY	At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.	At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.	At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.
MOLECULES ARE MOVING IN A REGULAR PATTERN	At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.	At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.	At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles.

Lw Pogil Kinetic Molecular Theory Answers

Brendan G. Carr



Lw Pogil Kinetic Molecular Theory Answers:

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