

Name: _____ Date: _____ Period: _____

pH Worksheet

1. What is pH a measure of?
2. What is the equation used for finding pH
3. What is the equation that relates to pH and pOH?
4. Complete the following table

$[H_3O^+]$	$[OH^-]$	pH	pOH	Acidic/Basic?
$1.0 \times 10^{-7} \text{ M}$	1×10^{-7}		5	
$2.4 \times 10^{-11} \text{ M}$		12.6	1.4	
	$5.62 \times 10^{-11} \text{ M}$	3.75	10.25	
$2.82 \times 10^{-9} \text{ M}$	$3.55 \times 10^{-6} \text{ M}$	8.55		

5. What would be the **pH** of each of the following:

- | | | | |
|------------------------------|-------|------------------------|-------|
| a) 0.0010 M HCl | _____ | g) 0.024 M HCl | _____ |
| b) 0.0010 M HNO ₃ | _____ | h) 0.075 M KOH | _____ |
| c) 0.010 M NaOH | _____ | i) 0.000034 M HCl | _____ |
| d) 0.0035 M HCl | _____ | j) 0.000000000001M HCl | _____ |
| e) 1.0 M HBr | _____ | | |
| f) 1.0 M KOH | _____ | | |

6. A 2.63 g NaOH are dissolved in 156 mL of solution. Determine the NaOH concentration & the pH.

7. List 3 strong acids and explain why these acids are considered strong acids.

8. List 3 weak acids and explain why these acids are considered weak acids.

9. List 2 strong bases and explain why these bases are considered strong bases.

10. List 1 weak base and explain why it is considered a weak base.

Instructional Fair Ph And Poh Continued

RC Schank



Instructional Fair Ph And Poh Continued:

The book delves into Instructional Fair Ph And Poh Continued. Instructional Fair Ph And Poh Continued is a vital topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Instructional Fair Ph And Poh Continued, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Instructional Fair Ph And Poh Continued
 - Chapter 2: Essential Elements of Instructional Fair Ph And Poh Continued
 - Chapter 3: Instructional Fair Ph And Poh Continued in Everyday Life
 - Chapter 4: Instructional Fair Ph And Poh Continued in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Instructional Fair Ph And Poh Continued. The first chapter will explore what Instructional Fair Ph And Poh Continued is, why Instructional Fair Ph And Poh Continued is vital, and how to effectively learn about Instructional Fair Ph And Poh Continued.
 3. In chapter 2, this book will delve into the foundational concepts of Instructional Fair Ph And Poh Continued. The second chapter will elucidate the essential principles that need to be understood to grasp Instructional Fair Ph And Poh Continued in its entirety.
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web solution $v_1 = 2.50 \text{ l}$ $v_2 = ?$ $t_1 = 273 \text{ K}$ $t_2 = 373 \text{ K}$ we will be applying charles law as pressure is constant here $v_1/t_1 = v_2/t_2$ or $v_2 = v_1 \cdot t_2 / t_1 = 2.50 \text{ l} \cdot 373 \text{ K} / 273 \text{ K} = 3.43 \text{ l}$ at 90 c a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml