

METRICS AND MEASUREMENT

Name _____

In the chemistry classroom and lab, the metric system of measurement is used, so it is important to be able to convert from one unit to another.

mega	kilo	hecto	deca	Basic Unit	deci	centi	milli	micro
(M)	(k)	(h)	(da)	gram (g)	(d)	(c)	(m)	(u)
1,000,000	1,000	100	10	liter (L)	.1	.01	.001	.000001
10^6	10^3	10^2	10^1	meter (m)	10^{-1}	10^{-2}	10^{-3}	10^{-6}

Factor Label Method

- Write the given number and unit.
- Set up a conversion factor (fraction used to convert one unit to another).
 - Place the given unit as denominator of conversion factor.
 - Place desired unit as numerator.
 - Place a "1" in front of the larger unit.
 - Determine the number of smaller units needed to make "1" of the larger unit.
- Cancel units. Solve the problem.

Example 1: 55 mm = _____ m

$$\frac{55 \text{ mm}}{1} \times \frac{1 \text{ m}}{1000 \text{ mm}} = 0.055 \text{ m}$$

Example 2: 88 km = _____ m

$$\frac{88 \text{ km}}{1} \times \frac{1000 \text{ m}}{1 \text{ km}} = 88,000 \text{ m}$$

Example 3: 7000 cm = _____ hm

$$\frac{7000 \text{ cm}}{1} \times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{1 \text{ hm}}{100 \text{ m}} = 0.7 \text{ hm}$$

Example 4: 8 dL = _____ dL

$$\frac{8 \text{ dL}}{1} \times \frac{10 \text{ L}}{1 \text{ dL}} \times \frac{10 \text{ dL}}{1 \text{ L}} = 800 \text{ dL}$$

The factor label method can be used to solve virtually any problem including changes in units. It is especially useful in making complex conversions dealing with concentrations and derived units.

Convert the following.

1. 35 mL = _____ dL

2. 950 g = _____ kg

3. 275 mm = _____ cm

4. 1,000 L = _____ kL

5. 1,000 mL = _____ L

6. 4,500 mg = _____ g

7. 25 cm = _____ mm

8. 0.005 kg = _____ dag

9. 0.075 m = _____ cm

10. 15 g = _____ mg

Instructional Fair Inc Density Answer Key

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Table of Contents Instructional Fair Inc Density Answer Key

1. Understanding the eBook Instructional Fair Inc Density Answer Key
 - The Rise of Digital Reading Instructional Fair Inc Density Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Instructional Fair Inc Density Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Instructional Fair Inc Density Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Instructional Fair Inc Density Answer Key
 - Personalized Recommendations
 - Instructional Fair Inc Density Answer Key User Reviews and Ratings

- Instructional Fair Inc Density Answer Key and Bestseller Lists
- 5. Accessing Instructional Fair Inc Density Answer Key Free and Paid eBooks
 - Instructional Fair Inc Density Answer Key Public Domain eBooks
 - Instructional Fair Inc Density Answer Key eBook Subscription Services
 - Instructional Fair Inc Density Answer Key Budget-Friendly Options
- 6. Navigating Instructional Fair Inc Density Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Instructional Fair Inc Density Answer Key Compatibility with Devices
 - Instructional Fair Inc Density Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Instructional Fair Inc Density Answer Key
 - Highlighting and Note-Taking Instructional Fair Inc Density Answer Key
 - Interactive Elements Instructional Fair Inc Density Answer Key
- 8. Staying Engaged with Instructional Fair Inc Density Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Instructional Fair Inc Density Answer Key
- 9. Balancing eBooks and Physical Books Instructional Fair Inc Density Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Instructional Fair Inc Density Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Instructional Fair Inc Density Answer Key
 - Setting Reading Goals Instructional Fair Inc Density Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Instructional Fair Inc Density Answer Key
 - Fact-Checking eBook Content of Instructional Fair Inc Density Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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