

# যশোর বোর্ড

১০। নিচের কোন যৌগটি পাকা কলায় বিদ্যমান?

- ☒ অ্যামাইল এসিটেট  
(খ) অক্টাইল এসিটেট  
(গ) ইথাইল বিউটাইরেট  
(ঘ) বেনজাইল এসিটেট

১১। আলোক সক্রিয় সমাপু হওয়ার জন্য অ্যালকেনের সর্বনিম্ন কার্বন সংখ্যা হবে—

- ☒ 7 (খ) 8  
(গ) 9 (ঘ) 10

১২। হির চাপে  $0^{\circ}\text{C}$  তাপমাত্রায়  $\text{O}_2$  গ্যাসের আয়তন 3.5L, হলে  $20^{\circ}\text{C}$  তাপমাত্রায় গ্যাসটির আয়তন হবে—

- (ক) 3.25L ☒ 3.76L  
(গ) 7.0L (ঘ) 8.0L

১৩। 0.15M NaOH দ্রবণের ppm ঘনমাত্রা কত?

- (ক) 4000 (খ) 5000  
(গ) 7000 ☒ 6000

১৪।  $2\text{FeCl}_3 + \text{SnCl}_2 \rightarrow 2\text{FeCl}_2 + \text{SnCl}_4$

- বিক্রিয়াটিতে জারিত হয়েছে কোনটি?  
(ক)  $\text{FeCl}_3$  (খ)  $\text{FeCl}_2$   
(গ)  $\text{SnCl}_4$  ☒  $\text{SnCl}_2$

১৫। কৃত্রিকভাবে সারফেন জটিলের বিসৃঙ্খতার মাপদণ্ড কোনটি?

- (ক) TDS (খ) COD  
☒ pH (ঘ) BOD

নিচের উদ্দীপকটি পড় এবং ১৬ ও ১৭ নং প্রশ্নের উত্তর দাও :

A ধাতুর ক্ষেত্রে  $E^{\circ}_{A/A^{2+}} = 0.88\text{V}$

B ধাতুর ক্ষেত্রে  $E^{\circ}_{B/B^{2+}} = -0.35\text{V}$

$E^{\circ}_{B^{2+}/B} = +0.35\text{V}$

১৬। উদ্দীপকের তথ্যসমূহের দ্বারা গঠিত কোষের জন্য—

- i.  $E^{\circ}_{\text{cell}} = E^{\circ}_{A/A^{2+}} + E^{\circ}_{B/B^{2+}}$   
ii.  $E^{\circ}_{\text{cell}} = E^{\circ}_{A/A^{2+}} + E^{\circ}_{B^{2+}/B}$   
iii.  $E^{\circ}_{\text{cell}} = E^{\circ}_{B^{2+}/B} - E^{\circ}_{A^{2+}/A}$

নিচের কোনটি সঠিক?

- (ক) i ও ii ☒ ii ও iii  
(গ) i ও iii (ঘ) i, ii ও iii

১৭। তড়িৎযান্ত্রিক দ্বারা গঠিত কোষের ক্ষেত্রে প্রযোজ্য হবে—

- (ক) A ধাতু ক্যাথোড হিসেবে ব্যবহৃত হয়  
☒ A ধাতু B অপেক্ষা অধিক সক্রিয়  
(গ) কোষটির ডায়গ্রাম হবে  $B/B^{2+} \parallel A^{2+}/A$   
(ঘ) কোষটির বিভব 0.53V

১৮।  $\text{CH}_3-\text{CH}=\text{CH}-\text{COOH}$  যৌগটির IUPAC নাম হলো—

- (ক) But-1-en-2-oic acid  
(খ) But-2-en-2-oic acid  
☒ But-2-en-1-oic acid  
(ঘ) Buten-1-oic acid

১৯। অম্লজলিক এসিডকে NaOH দ্বারা টাইট্রেশনের ক্ষেত্রে ব্যবহৃত নির্দেশকটি হলো—

- ☒ ফেনলফথ্যালিন  
(খ) মিথাইল রেড  
(গ) মিথাইল অরেঞ্জ  
(ঘ) থাইমল ব্লু

২০। কোনটি বিজারক পদার্থ?

- (ক)  $\text{KMnO}_4$  ☒  $\text{FeSO}_4$   
(গ)  $\text{H}_2\text{SO}_4$  (ঘ)  $\text{K}_2\text{Cr}_2\text{O}_7$

২১। নিচের কোনটির জন্য  $\frac{dV}{dT}$  প্রযোজ্য?

- (ক) মোলালিটি ☒ মোলালিটি  
(গ) নরমালিটি (ঘ) ফরমালিটি

২২। তড়িৎ বিশ্লেষণকালে কোনটি আগে জারিত হবে?

- (ক)  $\text{H}^+$  (খ)  $\text{Pb}^{2+}$   
☒  $\text{Cu}^{2+}$  (ঘ)  $\text{Na}^+$

২৩।  $\text{RCH}_2\text{CH}_2\text{X} \xrightarrow{\text{KCN}} \text{X} \xrightarrow[\text{H}^+]{\text{H}_2\text{O}} \text{Y}$

Y হলো—

- (ক) ক্রিটোন (খ) অ্যালকোহল  
(গ) অ্যালডিহাইড ☒ কার্বক্সিলিক এসিড

২৪।  $\text{CH}_3\text{CH}_2\text{Cl} \xrightarrow{\text{Na}} \text{B}$

তরু ইথার

$\xrightarrow{\text{KOH(aq)}} \text{C}$

B ও C উৎপাদদ্বয়ের সংকেত হল—

- ☒  $\text{CH}_3(\text{CH}_2)_2\text{CH}_3$  ও  $\text{CH}_3\text{CH}_2\text{OH}$   
(খ)  $\text{CH}_3\text{CH}_2\text{CH}_3$  ও  $\text{CH}_2=\text{CH}_2$   
(গ)  $\text{CH}_3(\text{CH}_2)_2\text{CH}_3$  ও  $\text{CH}_2=\text{CH}_2$   
(ঘ)  $\text{CH}_3\text{CH}_2\text{CH}_3$  ও  $\text{CH}_3\text{CH}_2\text{OH}$

২৫। কোনটি অগ্রতিসম অ্যালকিন?

- (ক)  $\text{CH}_2=\text{CH}_2$   
(খ)  $\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_3$   
☒  $\text{CH}_2=\text{CH}-\text{CH}_3$   
(ঘ)  $\text{ClCH}=\text{CHCl}$

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**C. M. Roebuck**



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web the legend says that if a person folds one thousand paper cranes out of paper he or she may be healed though each day sadako becomes weaker she decides to fold one thousand origami cranes her brother helps her by hanging the cranes from

the ceiling sadly sadako is unable to finish the project she passes away having made only 648

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web by takayuki ishii ten years after the atomic bomb was dropped on hiroshima sadako sasaki died as a result of the atomic bomb disease sadako s determination to fold one thousand paper cranes symbolizing her hope for peace and her courageous struggle with her illness inspired her classmates

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web after sadako s death in october of 1955 her classmates at school folded the remaining 356 cranes needed to reach one thousand after her funeral her class collected sadako s letters and journals and published them as a book which they called kokeshi

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