

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পটীকার প্রস্তুতকরণ

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$\text{মলের ভর, } m = 20 \text{ g}$$

$$\text{HCl এর আপেক্ষিক ভর,}$$

$$M = 36.5 \text{ g/mol}$$

$$\text{চলকের আয়তন, } V = 1 \text{ L}$$

$$\text{চলকের ঘনমাত্রা } S = 1$$

উদাহরণে উল্লিখিত N_2 -পদ্ধতিতে 10 mg এর মধ্যে কয় পরিমাণ ঘন ও V_1 -পদ্ধতিতে 0.2 g 2% $\left(\frac{W}{V}\right)$ HCl নমুনা নিয়ে কাজ করা হয়েছে।

যদিও N_2 -পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও V_1 -পদ্ধতিটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে মুদ্রণ ঘটে না।

$(CH_3NH_2 + H_2O \rightarrow CH_3NH_3^+ + OH^-)$ ফলে, কোনো কোনো ক্ষেত্রে কম কৃষিকার রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

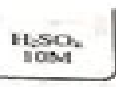
উদাহরণ



পাত্র-ক



পাত্র-খ



পাত্র-ঘ

১. উদাহরণে দু'ক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রকৃতিটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অংশেরে চলক থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার চলক তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ ও না প্রসঙ্গের উত্তর

৫. বিশ্লেষণের নিমিত্ত বীমের উপর রাইজার স্থাপন করলে বীমের প্রতি সারকে কতটা সোজা হয়, তাকে রাইজার খুবক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

৭. উদাহরণের চলকটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M চলক। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M চলক হলে পাত্র H_2SO_4 চলক। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঞ্জার প্রকৃতির অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঞ্জার প্রকৃতি অঙ্কন করে ব্যাখ্যা করা হলো:

প্রকৃতি	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান চলকে যে ক্ষয়কারী রাসায়নিক দ্রব্যকে মুদ্রণের সময় সন্নিবেশিত হয় ও ক্ষয়কারী পদার্থের হ্যাঞ্জারের প্রকৃতি নির্ধারিত হয়। এতে খুব খুঁচু হয় এবং ক্ষয়কারী পদার্থ পড়ে যায়।
উদাহরণ	পাত্র কঠিন এসিডসলফিউরিক (HCl , HNO_3 , H_2SO_4), পাত্র ক্ষয় ($NaOH$, KOH), H_2O_2 , অক্সিজেন লিকার প্রভৃতি চলক ইত্যাদি।
সংরক্ষণ পদ্ধতি	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র ক্ষয়কে, ক্ষয়কারী পদার্থের পিছনে রাখতে হবে।

৮. উদাহরণের চলকটি হলো 10M H_2SO_4 চলক এবং এর থেকে 1M ঘনমাত্রার চলক প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে হবে।

1ম চলকটির জন্য 100 ml অক্সিজেনমিত্রিক চলক ও এর টি হলো পিপিটি। সেখান থেকে অক্সিজেনমিত্রিক চলকের আয়তন 100 ml, তাই আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 চলক তৈরি করতে হবে। চলকের সন্নিবেশিত পাত্র হতে জানতে পারি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 চলকের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL।

আবার, পিপিটির সাহায্যে 10 ml প্রথম ঘনমাত্রার চলক নিয়ে তার সাথে অক্সিজেনমিত্রিক চলকে 90 ml পানি যোগ করলে আমাদের কামিষ্ঠ ঘনমাত্রার চলক পাওয়া যাবে।

উদাহরণ



চিত্র-A



চিত্র-B



চিত্র-C

১. অক্সিজেনমিত্রিক কাকে বলে?
২. সাদা খুবক K₂ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিত্রের হ্যাঞ্জারের উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক চলকটির ব্যবহারকালে খাখা কৃষ্টি হলে কবীর উপর সম্পর্কিত বিশ্লেষণ করে। ৪ ও না প্রসঙ্গের উত্তর

৫. অক্সিজেনমিত্রিক এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চেপক বা মুদ্রণ দ্রব্য উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে রক্ষা করে রাখার সঞ্চেপক করে।

৬. সামান্যত K₂ এর ঘন কখনো অসীম বা শূন্য হতে পারে না। উল্লিখিত ব্যাখ্যা করার জন্য নিম্নের উত্তরসূরী মাইক্রোপট বিশ্লেষণ করি-
 $A + B \rightarrow C + D$

Hsc Chemistry 1 First Chapter Notes

Karin Nielsen-Saines



Hsc Chemistry 1 First Chapter Notes:

Excel HSC Chemistry Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every single dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured This guide comes in a brand new format that makes even better use of your study time up to date coverage of the core topics plus 3 Option topics Industrial Chemistry Shipwrecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to section the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second hand and Investigations and Problem Solving %this way you will be able to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for word %this way you can easily match the Excel guide to the syllabus an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work hundreds of key concept questions with answers %questions that test your recall of knowledge in each chapter HSC type questions for every section in each chapter with clock icons to tell you how much time you will have to answer the questions in the HSC %this way you can test yourself on HSC type questions under HSC type time pressure an examiner maximiser feature ticks to show the mark distribution and answers to all HSC type questions %all you need to answer HSC type questions two sample HSC papers with an examiner maximiser feature plus answers %not one but two up to date sample papers the Excel syllabus summary notes a detachable section at the end of the guide where every single dot point of each chapter is summarised for you% a comprehensive and compact summary of the whole course in 32 pages

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