

GOVT 2305 You are to write a current event paper

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You are to write a current event paper, approximately 1000 words in length, consisting of a summary and analysis of a current policy issue having to do with American government and/or politics. You must **cite** a minimum of five references (sources) in the paper and include a **works cited page** at the end of the assignment to provide full bibliographic information for each of the references that was cited in the text of the paper.

To assist the student, a style guide for the paper is available for download.

Current event papers should be formatted according to MLA style, for papers with no title page (see the following link to access an online resource for MLA style: <https://owl.english.purdue.edu/owl/resource/747/08/>).

The current policy paper should have the following elements:

- Provide a thorough, but brief, summary of the political issue. An understanding of the issue should be developed by reading the articles uncovered during research for the assignment. The information provided in the summary will lay the foundation for you to react and respond to (analyze) the political issue. **Be sure to properly cite within the text (according to MLA style) any information that you obtained from one of your sources. Citations must be provided, even for paraphrased information.**
- Present your perspective on the political issue. Think critically about the topic discussed in the articles and clearly express your thoughts about it in writing. Consider eight standards for critical thinking (Paul and Elder, 2103): clarity, accuracy, precision,

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R Barnett



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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+ \rightarrow 2\text{AgCl}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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the complete ionic equation for this reaction is as follows equation 4 13 $2\text{Ag}^+ \text{aq} + 2\text{F}^- \text{aq} + 2\text{NH}_4^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7 \text{s} + 2\text{NH}_4^+ \text{aq} + 2\text{F}^- \text{aq}$ because two $\text{NH}_4^+ \text{aq}$

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