CHINE SOLL Lab 13 - Caloninetry

Name	DE WESTERN DE
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Complete the following density typing into the text box provided. The boxes will increase in size if additional space is necessary. Please he sure to save all your work as an MS Word document to submit properly.

Fact 1: Determining the Specific Rest Capacity of Aluminam

A. Dieta Table 1

Complete the Data Table. (7 pts):

ALUMENUM METAL (Trial 1)		
Fre-weighed aluminum metal mass (g)	20.03 g	
Initial temperature of cool water in the calorimeter (oC)	20.0%	
Institut temperature of aluminum metal (name as the boiling water in the pot) (oC)	100.0°C	
Pinal temperature of water/stummum metal in calorimeter (pC)	24 070	
ALUMENUM METAL (Trail 2)		
Fre-weighed sluminum metal mans (g) [Copy this from above since you are using the same metal sample for both trials].	20.03 g	
Initial temperature of cool water in the calorimeter (oC)	18.000	
Initial temperature of aluminum metal (mone as the bosling water in the pot) (o:C)	100 0°C	
Final temperature of waterfalamanum metal in calorimeter (pC)	21170	

It insert the following photo with your name and MEID showing in the photo-

Thoto A. Insert a photo of your calorimeter showing your aluminum metal submerged in the water. The photo should be taken from above. (6 pts)

- C. Calculations (Show all calculations with units on each number and the final answer rounded to the correct significant figures for full credit.)
 - Calculate the mass of the water by using the water's density (assume it is 1 00 g/mL) and the volume of the water. Remember, density = mass/volume. (4 pts)

$$D = \frac{M}{V} \rightarrow M = D = V \rightarrow M = \left[1.00 \frac{g}{ml}\right](75 mL) = 75 \text{ ga.kaminum} \text{ (Trial 1)}.$$

$$D = \frac{M}{V} \rightarrow M = D \times V \rightarrow M = \left[1.00 \frac{\alpha}{ml}\right](75 mL) = 75 \text{ gaAlaminum (Trud 2)}.$$

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AN Whitehead

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Table of Contents Gcc Chm130ll Nuclear Chemistry

- 1. Understanding the eBook Gcc Chm130ll Nuclear Chemistry
 - The Rise of Digital Reading Gcc Chm130ll Nuclear Chemistry
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Gcc Chm130ll Nuclear Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcc Chm130ll Nuclear Chemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Gcc Chm130ll Nuclear Chemistry
 - Personalized Recommendations
 - Gcc Chm130ll Nuclear Chemistry User Reviews and Ratings
 - Gcc Chm130ll Nuclear Chemistry and Bestseller Lists

- 5. Accessing Gcc Chm130ll Nuclear Chemistry Free and Paid eBooks
 - Gcc Chm130ll Nuclear Chemistry Public Domain eBooks
 - Gcc Chm130ll Nuclear Chemistry eBook Subscription Services
 - Gcc Chm130ll Nuclear Chemistry Budget-Friendly Options
- 6. Navigating Gcc Chm130ll Nuclear Chemistry eBook Formats
 - o ePub, PDF, MOBI, and More
 - Gcc Chm130ll Nuclear Chemistry Compatibility with Devices
 - Gcc Chm130ll Nuclear Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gcc Chm130ll Nuclear Chemistry
 - Highlighting and Note-Taking Gcc Chm130ll Nuclear Chemistry
 - Interactive Elements Gcc Chm130ll Nuclear Chemistry
- 8. Staying Engaged with Gcc Chm130ll Nuclear Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcc Chm130ll Nuclear Chemistry
- 9. Balancing eBooks and Physical Books Gcc Chm130ll Nuclear Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcc Chm130ll Nuclear Chemistry
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gcc Chm130ll Nuclear Chemistry
 - Setting Reading Goals Gcc Chm130ll Nuclear Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gcc Chm130ll Nuclear Chemistry
 - Fact-Checking eBook Content of Gcc Chm130ll Nuclear Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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