

Name Rocio Castro Period 4

## Chapter 3: Water and the Fitness of the Environment

### Concept 3.1 The polarity of water molecules results in hydrogen bonding

1. Study the water molecules at the right. On the central molecule, label oxygen (O) and hydrogen (H).
2. What is a *polar molecule*? Why is water considered polar?

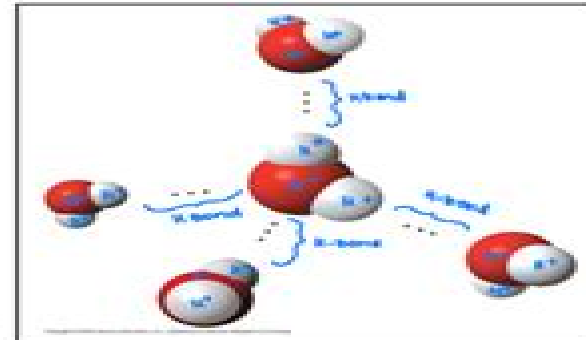
**Polar molecule**  
-unequal sharing of electrons  
-charge unevenly distributed

oxygen has 6 e<sup>-</sup> and 2 protons  
H = +1 O = -1  
 $2 \times +1 + 1 \times -1 = -2 + 1 = -1$

3. Now, add + and - signs to indicate the charged regions of each molecule. Then, indicate the hydrogen bonds.
4. Explain *hydrogen bonding*. How many hydrogen bonds can a single water molecule form?

4 H-bonds max per water molecule

H-bonding: noncovalent attraction between hydrogen & an electronegative atom nearby



### Concept 3.2 Four emergent properties of water contribute to Earth's fitness for life

Hydrogen bonding accounts for the unique properties of water. Let's look at several.

#### Cohesion

5. Distinguish between *cohesion* and *adhesion*.  
Cohesion: the H-bond tension between water molecules (intermolecular forces)  
Adhesion: the clinging of one substance to another (bonding between 2 unlike molecules) water H<sub>2</sub>O on side of cup after finish drinking
6. What is demonstrated when you see beads of water on a waxed car hood?  
cohesion & surface tension
7. Which property explains the ability of a water strider to walk on water?  
Surface tension

#### Moderation of Temperature

8. The calorie is a unit of heat. Define *calorie*.  
The amount of heat it takes to raise the temp of 1g of water by 1°
9. Water has high *specific heat*. What does this mean? How does water's specific heat compare to alcohol's?  
specific heat is the amount of heat that causes 1g of water to change temp by 1°C. water specific heat is 1 cal/gc. Alcohol requires less heat to change temp since it's specific heat is 0.6%
10. Explain how hydrogen bonding contributes to water's high specific heat.  
Hydrogen bonds are all over water molecules & it takes a lot of heat to break the H-bonds and a good amount of heat is released when H-bonds form  
\* Because H-bonds, water molecules move faster

# Fred And Theresa Holtzclaw Chapter 3vertebrates Answers

**Karin Nielsen-Saines**



**Fred And Theresa Holtzclaw Chapter 3vertebrates Answers:**

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In a digitally-driven earth wherever monitors reign supreme and instant interaction drowns out the subtleties of language, the profound techniques and psychological nuances concealed within words frequently get unheard. However, located within the pages of **Fred And Theresa Holtzclaw Chapter 3vertebrates Answers** a captivating literary value pulsing with organic emotions, lies an extraordinary journey waiting to be undertaken. Penned by a talented wordsmith, that enchanting opus invites visitors on an introspective trip, gently unraveling the veiled truths and profound influence resonating within the cloth of each and every word. Within the emotional depths of this moving review, we will embark upon a sincere exploration of the book's core styles, dissect its interesting publishing design, and fail to the effective resonance it evokes heavy within the recesses of readers' hearts.

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