

# Prelim Topic 1: The Chemical Earth

## 1. The living and non-living components of the Earth contain mixtures

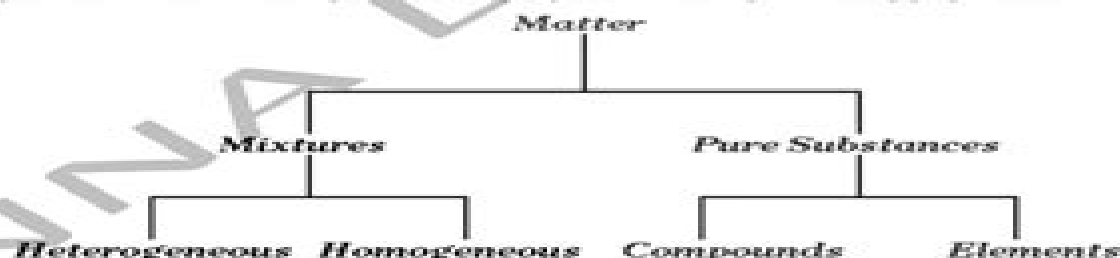
- ❖ Identify the difference between elements, compounds and mixtures in terms of particle theory.

### Particle theory

All matter are made up of tiny particles which are continually moving.

### Types of matter

1. Pure substance contains fixed compositions, fixed properties and cannot be decomposed by simple physical separation techniques.
  - a. **Elements** are composed of atoms or molecules
    - Simplest pure substance consisting of one type of atom
    - Cannot decompose
    - Exists as **molecules** (substances with two or more atoms of the same type chemically combined)
  - b. **Compounds** are composed of fixed numbers of atoms of different elements
    - Composed of two or more elements chemically bonded together
    - Fixed number of atoms of each component element
    - Decompose into component elements through chemical means
2. Mixture (Impure) have various particle types and compositions.
  - a. **Homogenous mixtures** have uniform composition and properties throughout eg. Air, water
  - b. **Heterogeneous mixtures** have variable composition and properties throughout eg. Blood, concrete
    - Components of mixture keep their own properties and separated by physical means eg. (filtering)



- ❖ Identify that the biosphere, lithosphere, hydrosphere and atmosphere contain examples of mixtures of elements and compounds

Biosphere		Major form	Common elements
	- the region on Earth where living organisms are found	Water	Hydrogen, oxygen
	-most living things are composed of cells which contain complex carbon compounds such as carbohydrates, fats and proteins	Calcium phosphate (bones)	Phosphorus, calcium
		Carbohydrates, fats	Carbon, hydrogen, oxygen
		Proteins	Carbon, hydrogen, oxygen, nitrogen, sulphur

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**Siddappa N.Byrareddy**



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## **Table of Contents Kiss Prelim Chemistry Topic 1 Chemical Earth Answers**

1. Understanding the eBook Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - The Rise of Digital Reading Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Kiss Prelim Chemistry Topic 1 Chemical Earth Answers

- Personalized Recommendations
- Kiss Prelim Chemistry Topic 1 Chemical Earth Answers User Reviews and Ratings
- Kiss Prelim Chemistry Topic 1 Chemical Earth Answers and Bestseller Lists
- 5. Accessing Kiss Prelim Chemistry Topic 1 Chemical Earth Answers Free and Paid eBooks
  - Kiss Prelim Chemistry Topic 1 Chemical Earth Answers Public Domain eBooks
  - Kiss Prelim Chemistry Topic 1 Chemical Earth Answers eBook Subscription Services
  - Kiss Prelim Chemistry Topic 1 Chemical Earth Answers Budget-Friendly Options
- 6. Navigating Kiss Prelim Chemistry Topic 1 Chemical Earth Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Kiss Prelim Chemistry Topic 1 Chemical Earth Answers Compatibility with Devices
  - Kiss Prelim Chemistry Topic 1 Chemical Earth Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Highlighting and Note-Taking Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Interactive Elements Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
- 8. Staying Engaged with Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
- 9. Balancing eBooks and Physical Books Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Setting Reading Goals Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Kiss Prelim Chemistry Topic 1 Chemical Earth Answers

- Fact-Checking eBook Content of Kiss Prelim Chemistry Topic 1 Chemical Earth Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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