#### **ANSWERS TO TOPIC 1**

#### **Review Questions**

1. 1	2. 3	3. 2
4. 4	5. 2	6. 2
7. 1	8. 3	9. 3

- 10. The plum pudding model pictures the atom with an even distribution of positive charges, with electrons throughout the atom. The planetary model pictures the nucleus at the center of the atom with electrons in orbits.
- 11. Rutherford fired positively charged alpha particles at gold feil. He expected them to pass through with little deflection. When some were greatly deflected, he reasoned that they must have been repelled by a massive positively charged particle, which was called the nucleus.
- The Rutherford model pictured the electrons in fixed, crititing positions. The wave-mechanical model pictures electrons as possessing both wave and particle properties.

13.	3	14.	3	15.	11.
16.	3	17.	4	1.8.	46
19.	2	20.	4	21.	122
22.	1	23.	1	24.	38
25.	3	26.	2	27.	22
28.	3	29.	3	30.	22
31.	3	32.	4	33.	40
34.	4	3.5.	1	36.	22
37.	1	38.	1	39.	23
40.	1	41.	2	42.	.3
43.	2	44.	4	45.	46
46.	3	47.	3	48.	44
49.	2	50.	3	51.	52
52.	2	53.	3	54.	1
55.	3	56.	3		
57.	Na 1s*2s*2p	*35m*			

	S 1s*2s*2p*3s*3p* Cl 1s*2s*2p*3s*3p* Ar 1s*2s*2p*3s*3p*					
. 61		59. 4	60. 1			
61.	1	62. 2	63. 4			
14.	4	65. 2	66. 2			
57.	2	66. 2	69. 1			

Mg 1e\*2e\*2p\*3e\*

Al 1s<sup>2</sup>2s<sup>2</sup>2p<sup>5</sup>3s<sup>3</sup>3p<sup>3</sup>

Si 1a\*2a\*2p\*3a\*3p\*

P 1s\*2s\*2o\*3s\*3o\*.

- 70. The mixture contains at least three parts. One solid remained on the filter paper. The other two parts consist of a liquid with a solid dissolved in it.
- If the calcium and sulfur can be separated by physical means, the substance is a mixture. If chemical means are required to separate them, it is a compound.
- The picture should show several of each type of melecule evenly mixed.
- 73. A mixture of iron and oxygen would have the iron present as a solid and the oxygen present as a gas. The iron would be attracted to a magnet. A glowing splint would ignite into a flame if placed into the oxygen gas. If the iron and oxygen were

combined as a compound, it would be a solid that is not attracted to a magnet.

- Pepperoni pizza is heterogeneous because individual parts can be seen.
- 75. A is a mixture of an element and a compound; one molecule centains identical atoms and the other one contains different types of atoms. B is a compound; all particles are identical and are made up of two different types of atoms. C is an element; all molecules are identical and each is made up of two identical atoms. D is a mixture of an element and two different compounds; one molecule contains identical atoms and the other two contain different types of atoms, but they differ from each other.
- Container D is homogeneous because the particles are mixed on the molecular level.
- The two substances are different materials because their composition varies.

#### **Regents Practice Questions**

Part A		
F. 3.	2. 1	3. 1
4. 3	5. 1	6. 4
7. 2	8. 3	9. 3
10. 2	11. 3	12. 1
13. 2	14. 3	
Part B		
15. 4	16. 3	17. 1
18. 3	19. 1	20. 3
21. 1	22. 3	23. 2
24. 4		

#### Point C

- 25. Because the two nuclei have different numbers of protons, they have different amounts of attraction for electrons. As a result, the amounts of energy separating the energy levels differ for the two elements.
- The samples are not the same substance because the percent of each element differs from sample to sample.
- 27. Because he assumed the atom to have an even distribution of positive charges, he did not expect a concentration of positive charges that would strongly deflect or bounce back the alpha particles.
- 28. The spectra are based on the elements present in the compound. If some lines matched and some didn't, the compounds have a least one element in common and at least one differing element.
- Column 1: 1, 2, 1; column 2: K, 39, 19; column 3: 92, 146, 92
- **30.** (0.9778)(12) + (0.0222)(13) 12.02
- 31. They contain different numbers of neutrons.
- **32.**  $(19.99 \times .909) + (20.99 \times 0.03) + (21.99 \times 0.88)$
- 33. 20
- $(12.00 \times 9.93) + (13.00 \times 1.07)$

#### 1.00

35. Both isotopes contain 6 protons but different numbers of neutrons. Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 protons and 7 neutrons.

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the hitchhiker's guide to the galaxy film wikipedia - Apr 11 2023

web plot one thursday morning arthur dent discovers that his house is to be immediately demolished to make way for a bypass he tries delaying the bulldozers by lying down in front of them ford prefect a friend of arthur s convinces him to go to a pub with him

the hitchhiker's guide to the galaxy radio series wikipedia - Nov 06 2022

web the hitchhiker's guide to the galaxy is a science fiction comedy radio series primarily written by douglas adams it was originally broadcast in the united kingdom by bbc radio 4 in 1978 and afterwards the bbc world service national public radio in the us and cbc radio in canada the series was the first radio comedy programme to be produced in

#### the hitchhiker's guide to the galaxy study guide litcharts - Sep 04 2022

web the hitchhiker's guide to the galaxy is the first book in douglas adams's series of five novels tracing arthur dent's journey through space the second installation is called the restaurant at the end of the universe and it picks up where the hitchhiker's guide leaves off following arthur and his friends as they leave magrathea and head for a quick bite the hitchhiker's guide to the galaxy nature - Dec 07 2022

web oct 2 2019 don't panic the plot of the hitchhiker's guide centres on permanently bemused human protagonist arthur dent who wanders the universe after the destruction of earth with alien travel writer ford