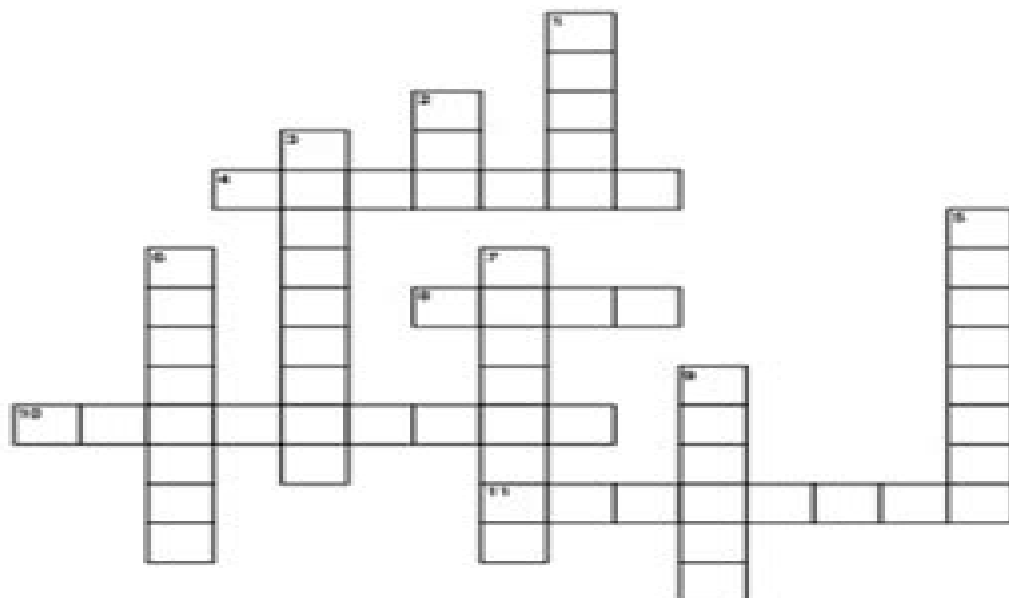


# Chemical Bonding



## Across

- [4] the number of electron needed to make a full outer shell
- [6] The valency of Carbon
- [10] two of these bond to form a covalent bond
- [11] An ion that has gained electrons

## Down

- [1] A bond formed when electrons are given away or received
- [2] The valency of Chlorine
- [3] The man who discovered the law of conservation of mass
- [5] An ion the has lost electrons
- [6] What we call an equation with equal numbers of elements on both sides
- [7] A bond formed by sharing electrons
- [9] Elements found on the left side of the periodic table

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**LM Reder**



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