

Chemistry – Mock Test Paper

[Time: $1\frac{1}{2}$ hrs]

[M. Marks : 80]

Answers to this paper must be written on the paper provided separately.

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt any four questions from Section II.

Section I (40 Marks)

Attempt all questions from this section

Question 1.

(a) Choose the most appropriate answer.

- (i) The element with largest size in second period is :
 A. Lithium B. Oxygen C. Fluorine D. Carbon
- (ii) Which is the element which belongs to group 13 from :
 A. Beryllium B. Boron C. Aluminium D. Carbon
- (iii) The most strong alkali is
 A. Lithium hydroxide B. Potassium hydroxide
 C. Sodium hydroxide D. Rubidium hydroxide
- (iv) Lanthanides is a group of elements outside the Periodic Table having atomic number from
 A. 57 to 71 B. 89 to 103 C. 72 to 86 D. 104 to 118
- (v) The elements with lowest first ionisation energy is
 A. Sodium B. Caesium C. Barium D. Magnesium
- (vi) When equal volumes of hydrogen gas and chlorine gas are exposed to diffused sunlight the reaction :
 A. does not take place B. takes place at moderate speed
 C. is explosive in nature D. none of the above.
- (vii) In laboratory ammonia gas prepared by heating a mixture of :
 A. NH_4Cl and $\text{Ca}(\text{OH})_2$ B. $(\text{NH}_4)_2\text{SO}_4$ and $\text{Ca}(\text{OH})_2$
 C. NH_4NO_3 and KOH D. NH_4NO_3 and $\text{Ca}(\text{OH})_2$
- (viii) The compound which will liberate carbon dioxide on treating with dilute sulphuric acid is :
 A. MgCl_2 B. Na_2SO_4 C. Na_2CO_3 D. Na_2SO_4
- (ix) Which is not a member of alkanolic series?
 A. HCOOH B. $\text{C}_2\text{H}_5\text{COOH}$ C. HCHO D. CH_3COOH
- (x) Ethyne can be converted into ethyne ozonide by treating with :
 A. Hydrogen peroxide B. Ozone
 C. Oxygen D. Conc. HNO_3

[10]

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Table of Contents Grade 10 Chemistry Paper November Exam

1. Understanding the eBook Grade 10 Chemistry Paper November Exam
 - The Rise of Digital Reading Grade 10 Chemistry Paper November Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 10 Chemistry Paper November Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 10 Chemistry Paper November Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 10 Chemistry Paper November Exam
 - Personalized Recommendations
 - Grade 10 Chemistry Paper November Exam User Reviews and Ratings
 - Grade 10 Chemistry Paper November Exam and Bestseller Lists
5. Accessing Grade 10 Chemistry Paper November Exam Free and Paid eBooks
 - Grade 10 Chemistry Paper November Exam Public Domain eBooks

- Grade 10 Chemistry Paper November Exam eBook Subscription Services
- Grade 10 Chemistry Paper November Exam Budget-Friendly Options
- 6. Navigating Grade 10 Chemistry Paper November Exam eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 10 Chemistry Paper November Exam Compatibility with Devices
 - Grade 10 Chemistry Paper November Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 10 Chemistry Paper November Exam
 - Highlighting and Note-Taking Grade 10 Chemistry Paper November Exam
 - Interactive Elements Grade 10 Chemistry Paper November Exam
- 8. Staying Engaged with Grade 10 Chemistry Paper November Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 10 Chemistry Paper November Exam
- 9. Balancing eBooks and Physical Books Grade 10 Chemistry Paper November Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 10 Chemistry Paper November Exam
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 10 Chemistry Paper November Exam
 - Setting Reading Goals Grade 10 Chemistry Paper November Exam
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 10 Chemistry Paper November Exam
 - Fact-Checking eBook Content of Grade 10 Chemistry Paper November Exam
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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