

13.2**THE NATURE OF LIQUIDS****Section Review****Objectives**

- Identify factors that determine physical properties of a liquid
- Define *evaporation* in terms of kinetic energy
- Describe the equilibrium between a liquid and its vapor
- Identify the conditions under which boiling occurs

Vocabulary

- vaporization
- evaporation
- vapor pressure
- boiling point
- normal boiling point

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Liquids are much 1 than gases. Liquids and solids are known as 2 states of matter. The conversion of a liquid to a gas or vapor is called 3. When a liquid that is not 4 changes to a gas, the process is called evaporation. A liquid evaporates faster when heated; however, evaporation itself is a 5 process. When a partially filled container of liquid is sealed, some of the particles at the 6 of the liquid vaporize. These particles collide with the walls of the container, producing a force called 7. The vapor pressure of a liquid can be determined by a device called a 8. A liquid boils when its 9 equals the external pressure. The normal boiling point of a liquid is the temperature at which the vapor pressure is equal to 10.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 11. Particles in a liquid don't have enough kinetic energy to overcome the attractive forces between them and vaporize.

Nature Of Liquids Section Review Answers

Peter Wasserscheid, Thomas Welton



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Solid State Physics ,1978-10-27 Solid State Physics *Excitations in a Bose-condensed Liquid* Allan Griffin,1993-08-19 This volume gives an up to date systematic account of the microscopic theory of Bose condensed fluids developed since the late 1950s In contrast to the usual phenomenological discussions of superfluid ^4He the present treatment is built on the pivotal role of the Bose broken symmetry and a Bose condensate The many body formalism is developed with emphasis on the one and two particle Green s functions and their relation to the density response function These are all coupled together by the Bose broken symmetry which provides the basis for understanding the elementary excitations and response functions in the hydrodynamic and collisionless regions It also explains the difference between excitations in the superfluid and normal phases Chapter 4 gives the first critical assessment of the experimental evidence for a Bose condensate in liquid ^4He based on high momentum neutron scattering data *Ionic Liquids in Synthesis* Peter Wasserscheid,Thomas Welton,2008-06-25

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