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### Frage 1: Wasser-Wasserdampf System

Wie kann man feststellen, ob das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen

- Erhöht sich die Temperatur des Systems, wenn man das Volumen des Systems konstant hält?
- Erhöht sich die Temperatur des Systems, wenn man das Volumen des Systems konstant hält?

Wenn man das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen feststellt, dann kann man feststellen, ob das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen

- Erhöht sich die Temperatur des Systems, wenn man das Volumen des Systems konstant hält?
- Erhöht sich die Temperatur des Systems, wenn man das Volumen des Systems konstant hält?

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ist das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen festzustellen, dann kann man feststellen, ob das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen

Wenn man das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen feststellt, dann kann man feststellen, ob das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen

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- Erhöht sich die Temperatur des Systems, wenn man das Volumen des Systems konstant hält?

Es ist nicht möglich, das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen festzustellen, dann kann man feststellen, ob das thermodynamische Gleichgewicht bei konstantem (und) konstantem Volumen

Table 1: The thermodynamic properties of water and steam at standard atmospheric pressure.

Phase	Formula	Temperature (°C)	Enthalpy (kJ/kg)
Ice (solid)	$H_2O$	0	0
Water (liquid)	$H_2O$	100	419
Steam (gas)	$H_2O$	100	2260
Ice (solid)	$H_2O$	0	0
Water (liquid)	$H_2O$	100	419
Steam (gas)	$H_2O$	100	2260

# Modeling Chemistry U5 Ws1 V1 1 Key

**M Carnoy**



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