

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that our "paper" writer is a chemist who has to write out what products may be made. What are the possibilities and how do you find them? The next 4 problems cover typical questions.

1. Chlorine reacts with bromine in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that some decomposition for products produces chlorine, Br_2 say, and 1.0 moles of oxygen gas.

Reactants	1.0 Br_2	1.0 Cl_2	0	0
Products	0	0	0	0
Change	1.0	1.0	0	0
Initial	0	0	0	0

1.0 moles of Br_2 and 1.0 moles of Cl_2

2. How much a displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How much of hydrochloric acid is required to produce 1.0 mole of H_2 gas?

Reactants	1.0 Mg	1.0 HCl	0	0
Products	0	0	0	0
Change	1.0	1.0	0	0
Initial	0	0	0	0

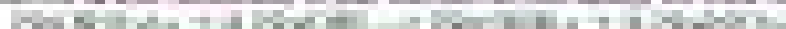
1.0 moles of Mg and 1.0 moles of HCl

3. Chlorine, Cl_2 , reacts with hydrogen gas to produce hydrogen chloride gas. How much gas is required to produce 1.0 mole of hydrogen gas by displacement of chlorine from sodium?

Reactants	1.0 Cl_2	1.0 H_2	0	0
Products	0	0	0	0
Change	1.0	1.0	0	0
Initial	0	0	0	0

1.0 moles of Cl_2 and 1.0 moles of H_2

4. Chlorine reacts with sodium to produce sodium chloride. How much sodium is required to produce 1.0 mole of sodium chloride?



1.0 moles of Na and 0.5 moles of Cl_2

Reactants	1.0 Na	0.5 Cl_2	0	0
Products	0	0	0	0
Change	1.0	0.5	0	0
Initial	0	0	0	0

1.0 moles of Na and 0.5 moles of Cl_2

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