

$$PV = nRT$$

Gas Stoichiometry Worksheet

$$PV = \frac{m}{M}RT \quad \frac{m}{V} = \frac{PM}{RT}$$

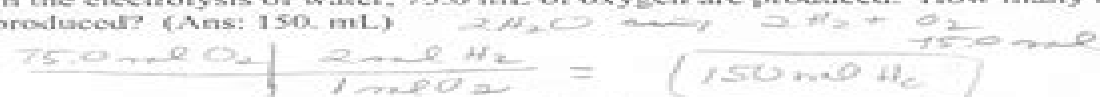
Name: Key

Date: _____

Period: _____

Volume - Volume

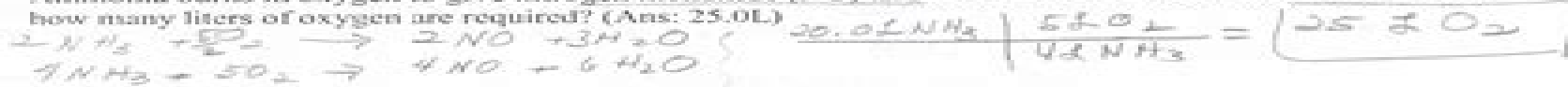
1. In the electrolysis of water, 75.0 mL of oxygen are produced. How many milliliters of hydrogen are produced? (Ans: 150. mL)



2. How many liters of hydrogen are required to react with excess nitrogen to produce 50.0 L ammonia (NH₃)? (Ans: 75.0 L)



3. Ammonia burns in oxygen to give nitrogen monoxide (NO) and water. If 20.0 L of ammonia are burned, how many liters of oxygen are required? (Ans: 25.0 L)

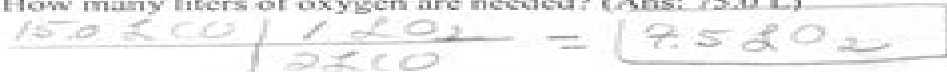


4. Carbon monoxide burns in oxygen and forms carbon dioxide.

- a. How many liters of carbon dioxide are produced from 15.0 L of CO? (Ans: 15.0 L)



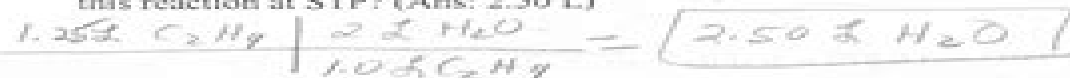
- b. How many liters of oxygen are needed? (Ans: 7.5 L)



5. If 400 mL of hydrogen and 400 mL of oxygen are mixed and ignited, what volume of oxygen remains uncombined? (Ans: 200 mL)



6. Ethylene burns in oxygen to form carbon dioxide and water vapor in the following reaction: $C_2H_4(g) + 3O_2(g) \rightarrow 2CO_2(g) + 2H_2O(g)$. How many liters of water can be formed if 1.25 liters of ethylene are consumed in this reaction at STP? (Ans: 2.50 L)



Mass - Volume



1. How many grams of sodium are needed to release 4.0 L of hydrogen from water? (Ans: 8.21 g)



2. When 130 g of zinc react with excess hydrochloric acid, what volume of hydrogen is produced? (Ans: 44.8 L)



Mixed Stoichiometry Problems Key

Jin-Ying Zhang



Mixed Stoichiometry Problems Key:

The book delves into Mixed Stoichiometry Problems Key. Mixed Stoichiometry Problems Key is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Mixed Stoichiometry Problems Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Mixed Stoichiometry Problems Key
- Chapter 2: Essential Elements of Mixed Stoichiometry Problems Key
- Chapter 3: Mixed Stoichiometry Problems Key in Everyday Life
- Chapter 4: Mixed Stoichiometry Problems Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Mixed Stoichiometry Problems Key. This chapter will explore what Mixed Stoichiometry Problems Key is, why Mixed Stoichiometry Problems Key is vital, and how to effectively learn about Mixed Stoichiometry Problems Key.
3. In chapter 2, the author will delve into the foundational concepts of Mixed Stoichiometry Problems Key. The second chapter will elucidate the essential principles that must be understood to grasp Mixed Stoichiometry Problems Key in its entirety.
4. In chapter 3, this book will examine the practical applications of Mixed Stoichiometry Problems Key in daily life. This chapter will showcase real-world examples of how Mixed Stoichiometry Problems Key can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Mixed Stoichiometry Problems Key in specific contexts. This chapter will explore how Mixed Stoichiometry Problems Key is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Mixed Stoichiometry Problems Key. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Mixed Stoichiometry Problems Key.

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