Grade	12 Paper 2 (Chemistry)	MEMO	September 2014	
1.1	B			
1.2	A			
1.4	8			
1.5	A			
1.6	A -			
1.7	D **			
1.8	D			
1.9	c			
1.10	D ·		$(2 \times 10 = 20)$	
2.1	H - C - C - C - C - C - C - C - C - C -	—	-H (3)	
2.2	They are volatile and evaporate • quickly.		(1)	
2.4	hydrolysis/elimination/esterification/conden	302,524 / 90235		
2.4	hydrolysis/elimination/esterification/conden	sation (any	one) (1)	
2.5	heat * and sulphuric acid (H ₂ SO ₄) *		(2)	
3.1	Methane is a non-polar substance → and is the and contains and –OH group → which allows			
3.2	Methane and Methanal		(2)	
3.3	As the mass of the carboxylic acid increases, the boiling point of the carboxylic acids also increases. This is because larger molecules have more intermolecular forces. than smaller molecules.			
4.1.1	Hydration		(1)	
4.1.2	No [✓] because it needs a catalyst and heat [✓]	to take place.	(2)	
4.1.3	propan ~ -1-ol ~		(2)	
4.1.4	primary *		(1)	
4.1.5	No ♥, also propan-2-ol ♥		(2)	
4.2.1	hydrohalogenation		(1)	
4.2.2	no water present		(1)	

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Tobias Bleicker

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