

## CHAPTER 7 PRACTICE TEST: ATOMIC STRUCTURE &amp; PERIODIC TRENDS- KEY

1) When a solution of sodium chloride is vaporized in a flame, the color of the flame is:

- A) Blue      **B) Yellow**      C) Green      D) Violet      E) White

2) Electromagnetic radiation travels through vacuum at a speed of \_\_\_\_\_ m/s.

- A) 186,000      **C)  $3.00 \times 10^8$**       E) It depends on  $\lambda$   
B) 125      D) 10,000

3) The wavelength of light that has a frequency of  $1.20 \times 10^{13} \text{ s}^{-1}$  is \_\_\_\_\_ m.

- A) 25.0      **B)  $2.50 \times 10^{-5}$**       C) 0.0400      E) 2.5  
D) 12.0

GIVEN:  $\nu = 1.20 \times 10^{13} \text{ s}^{-1}$

UNKNOWN:  $\lambda$

CONNECTION:  $c = \lambda \nu$

SUBSTITUTE & SOLVE:  $\lambda = c/\nu$

$$\lambda = \frac{3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{1.20 \times 10^{13} \text{ s}^{-1}}$$

$$\lambda = 2.50 \times 10^{-5} \text{ m}$$

4) Ham radio operators often broadcast on the 6-meter band. The frequency of this EM radiation is \_\_\_\_\_ MHz.

- A) 500      B) 200      **C) 50**      D) 20      E) 2.0

GIVEN:  $\lambda = 6 \text{ m}$

UNKNOWN:  $\nu$  (Must convert answer to MHz!)

CONNECTION:  $c = \lambda \nu$

SUBSTITUTE & SOLVE:  $\nu = c/\lambda$

$$\nu = \frac{3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{6 \text{ m}}$$

$$\nu = 0.5 \times 10^8 \text{ Hz} \times \frac{1 \text{ MHz}}{10^6 \text{ Hz}} = 50 \text{ MHz}$$

5) The energy of a photon of light is \_\_\_\_\_ proportional to its frequency and \_\_\_\_\_ proportional to its wavelength.

- A) directly, directly      C) inversely, directly      E) indirectly, not  
B) inversely, inversely      **D) directly, inversely**

6) The energy of a photon that has a wavelength of 12.3 nm is \_\_\_\_\_ J.

- A)  $1.51 \times 10^{-17}$       B)  $4.42 \times 10^{-23}$       C)  $1.99 \times 10^{-23}$       D)  $2.72 \times 10^{-19}$       **E)  $1.62 \times 10^{-17}$**

7) The wavelength of a photon that has an energy of  $5.25 \times 10^{-19} \text{ J}$  is \_\_\_\_\_ m.

- A)  $3.79 \times 10^{-7}$**       B)  $2.64 \times 10^6$       C)  $2.38 \times 10^{23}$       D)  $4.21 \times 10^{-24}$       E)  $3.79 \times 10^7$

8) A mole of red photons of wavelength 725 nm has \_\_\_\_\_ kJ of energy.

- A)  $2.74 \times 10^{-19}$       B)  $4.56 \times 10^{-19}$       C)  $6.05 \times 10^{-7}$       **D) 165**      E) 227

GIVENS:  $\lambda = 725 \text{ nm}$  (must convert to m before using formulas)

$$\lambda = 725 \text{ nm} \times \frac{1 \text{ m}}{10^9 \text{ nm}} = 7.25 \times 10^{-7} \text{ m}$$

# Modeling Chemistry Unit 6 Test

**Lingsheng Yao**



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Jespersen, Pamela Kerrigan, 2023-07-04 Always study with the most up to date prep Look for AP Chemistry Premium 2025 Prep Book with 6 Practice Tests Comprehensive Review Online Practice ISBN 9781506291802 on sale July 2 2024 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entities included with the product

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Courtney A. Young, 2019-02-01 This landmark publication distills the body of knowledge that characterizes mineral processing and extractive metallurgy as disciplinary fields It will inspire and inform current and future generations of minerals and metallurgy professionals Mineral processing and extractive metallurgy are atypical disciplines requiring a combination of knowledge experience and art Investing in this trove of valuable information is a must for all those involved in the industry students engineers mill managers and operators More than 192 internationally recognized experts have contributed to the handbook's 128 thought provoking chapters that examine nearly every aspect of mineral processing and extractive metallurgy This inclusive reference addresses the magnitude of traditional industry topics and also addresses the new technologies and important cultural and social issues that are important today Contents Mineral Characterization and Analysis Management and Reporting Comminution Classification and Washing Transport and Storage Physical Separations Flotation Solid and Liquid Separation Disposal Hydrometallurgy Pyrometallurgy Processing of Selected Metals Minerals and Materials

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### **Selected**

**Water Resources Abstracts**, 1988-04

**Heath Science Connections 9** Jack L. Candido, Edward S. James, Ronald E.

Phillips, 1987

**Progress in Modeling and Simulation of Batteries** John Turner, 2016-06-15 Modeling and simulation of

batteries in conjunction with theory and experiment are important research tools that offer opportunities for advancement of technologies that are critical to electric motors The development of data from the application of these tools can provide the basis for managerial and technical decision making Together these will continue to transform batteries for electric vehicles This collection of nine papers presents the modeling and simulation of batteries and the continuing contribution being made to this impressive progress including topics that cover Thermal behavior and characteristics Battery management system design and analysis Moderately high fidelity 3D capabilities Optimization Techniques and Durability As electric vehicles continue to gain interest from manufacturers and consumers alike improvements in economy and affordability as well as adoption of alternative fuel sources to meet government mandates are driving battery research and development Progress in modeling and simulation will continue to contribute to battery improvements that deliver increased power energy storage and durability to further enhance the appeal of electric vehicles

**The City Record** New York (N.Y.),1916 *Energy Research Abstracts* ,1992 Technical Abstract Bulletin , *El-Hi Textbooks in Print* ,1981 **Research in Education** ,1972 *Water Treatment Unit Processes* David W. Hendricks,2018-10-03 The unit process approach common in the field of chemical engineering was introduced about 1962 to the field of environmental engineering An understanding of unit processes is the foundation for continued learning and for designing treatment systems The time is ripe for a new textbook that delineates the role of unit process principles in environmental engineering Suitable for a two semester course *Water Treatment Unit Processes Physical and Chemical* provides the grounding in the underlying principles of each unit process that students need in order to link theory to practice Bridging the gap between scientific principles and engineering practice the book covers approaches that are common to all unit processes as well as principles that characterize each unit process Integrating theory into algorithms for practice Professor Hendricks emphasizes the fundamentals using simple explanations and avoiding models that are too complex mathematically allowing students to assimilate principles without getting sidelined by excess calculations Applications of unit processes principles are illustrated by example problems in each chapter Student problems are provided at the end of each chapter the solutions manual can be downloaded from the CRC Press Web site Excel spreadsheets are integrated into the text as tables designated by a CD prefix Certain spreadsheets illustrate the idea of scenarios that emphasize the idea that design solutions depend upon assumptions and the interactions between design variables The spreadsheets can be downloaded from the CRC web site The book has been designed so that each unit process topic is self contained with sidebars and examples throughout the text Each chapter has subheadings so that students can scan the pages and identify important topics with little effort Problems references and a glossary are found at the end of each chapter Most chapters contain downloadable Excel spreadsheets integrated into the text and appendices with additional information Appendices at the end of the book provide useful reference material on various topics that support the text This design allows students at different levels to easily navigate through the book and professors to assign pertinent sections in

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