

Name _____

Date _____ Per _____

Mixed Practice: Names and Formulas

First determine whether each of the following compounds is *ionic* ($M + NM$), *polyatomic* (3 or more atoms), or *covalent* (2 NM's). Then write the name for each compound.

1) NO_2 _____

2) NaBr _____

3) SiO_2 _____

4) P_2Br_4 _____

5) FeSO_4 _____

6) SF_6 _____

7) Li_2S _____

8) MgBr_2 _____

9) N_2S _____

10) $\text{Be}(\text{OH})_2$ _____

11) SO_3 _____

12) Cu_2S _____

13) BF_3 _____

14) I_2 _____

15) $\text{Pb}_3(\text{PO}_3)_2$ _____

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SJ Ball



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