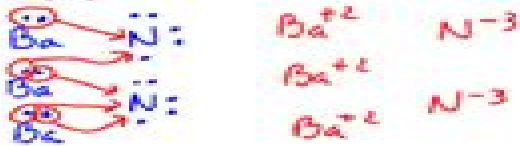


Draw the dot diagrams for each element in the compound

1. NaCl  $\text{Na}^+ \text{Cl}^-$	7. Li_2O  $\text{Li}^+ \text{O}^{2-} \text{Li}^+$
2. CaO  $\text{Ca}^{+2} \text{O}^{-2}$	8. Ag_3P  $\text{Ag}^+ \text{P}^{-3} \text{Ag}^+ \text{Ag}^+$
3. AlN  $\text{Al}^{+3} \text{N}^{-3}$	9. Ba_3N_2  $\text{Ba}^{+2} \text{N}^{-3} \text{Ba}^{+2} \text{Ba}^{+2} \text{N}^{-3}$
4. MgS  $\text{Mg}^{+2} \text{S}^{-2}$	10. Al_2O_3  $\text{Al}^{+3} \text{O}^{-2} \text{Al}^{+3} \text{O}^{-2} \text{O}^{-2}$
5. FeCl_2  $\text{Fe}^{+2} \text{Cl}^- \text{Cl}^-$	11. ZnI_2  $\text{Zn}^{+2} \text{I}^- \text{I}^-$
6. CoF_3  $\text{Co}^{+3} \text{F}^- \text{F}^- \text{F}^-$	12. Au_2S_3  $\text{Au}^{+3} \text{S}^{-2} \text{Au}^{+3} \text{S}^{-2} \text{S}^{-2}$

Modeling Lewis Dot Structures Answer Key

Christian Drosten



Modeling Lewis Dot Structures Answer Key:

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