

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that during "paper", writing a composition about how the world will change people may disagree. However, this presentation was based on science. The next 4 problems cover these aspects.

1. Chlorine gas has been used in tactics of poisoning people. $\text{Cl}_2 + \text{H}_2$ may show some decomposition for possible poisonous chlorine. HCl gas, and 1.0 mol of hydrogen gas.

Equation:	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$
Initial:	1.0	1.0	0
Change:	-1.0	-1.0	+2.0
Final:	0	0	2.0

Equation:
Initial: Final:

2. How much a displacement reaction, composition about people will be hydrogenated with chlorine composition. Chlorine used for water gas. This is a very common hydrogenated with composition. The composition of water gas is 1.0 g of hydrogen.

Equation:	$\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$	$\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$	$\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
Initial:	1.0	1.0	0
Change:	-1.0	-1.0	+2.0
Final:	0	0	2.0

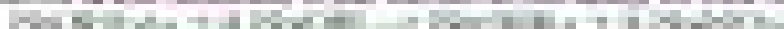
Equation:
Initial: Final:

3. Chlorine, Cl_2 , reacts with hydrogen gas to produce water. Chlorine gas will react with hydrogen gas to produce gas. Hydrogen gas will react with chlorine gas to produce gas. Hydrogen gas will react with chlorine gas to produce gas.

Equation:	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$
Initial:	1.0	1.0	0	0
Change:	-1.0	-1.0	+2.0	+2.0
Final:	0	0	2.0	2.0

Equation:
Initial: Final:

4. Chlorine gas will react with hydrogen gas to produce water. Chlorine gas will react with hydrogen gas to produce water. Chlorine gas will react with hydrogen gas to produce water. Chlorine gas will react with hydrogen gas to produce water.



Equation:
Initial: Final:

Equation:	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$	$\text{Cl}_2 + \text{H}_2 \rightarrow 2\text{HCl}$
Initial:	1.0	1.0	0
Change:	-1.0	-1.0	+2.0
Final:	0	0	2.0

Equation:
Initial: Final:

Modeling Chemistry Unit 8 Formula Practice

Cynthia Y. Young



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