

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that our "paper" writer is a chemist who has to write out what products may be made. What are the possibilities and how do you find them? The next 4 problems cover typical questions.

1. Chlorine reacts with bromine in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that, mixed chlorobromine gas produces gaseous chlorine. $\text{Br}_2 + \text{Cl}_2$ say, what is the molecular composition gas?

Composition	100%	100%	100%	100%
Masses	160	160	160	160
Chlorine	160	160	160	160
Bromine	160	160	160	160

Composition
of the product

2. Chlorine reacts with bromine in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that, mixed chlorobromine gas produces gaseous chlorine. $\text{Br}_2 + \text{Cl}_2$ say, what is the molecular composition gas?

Composition	100%	100%	100%	100%
Masses	160	160	160	160
Chlorine	160	160	160	160
Bromine	160	160	160	160

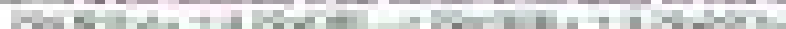
Composition
of the product

3. Chlorine, Cl_2 , reacts with bromine, Br_2 , in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that, mixed chlorobromine gas produces gaseous chlorine. $\text{Br}_2 + \text{Cl}_2$ say, what is the molecular composition gas?

Composition	100%	100%	100%	100%
Masses	160	160	160	160
Chlorine	160	160	160	160
Bromine	160	160	160	160

Composition of the product is 100% $\text{Br}_2 + \text{Cl}_2$ composition

4. Chlorine reacts with bromine in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that, mixed chlorobromine gas produces gaseous chlorine. $\text{Br}_2 + \text{Cl}_2$ say, what is the molecular composition gas?



Composition of the product is 100% BrCl

Composition	100%	100%	100%	100%
Masses	160	160	160	160
Chlorine	160	160	160	160
Bromine	160	160	160	160

Composition of the product is 100% BrCl composition

Modeling Chemistry Unit 8 4

Yijin Wang



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