

Chemistry Unit 7 Worksheet 4

Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedures outlined in class. Be sure to circle your final answer.

- Calculate the number of moles of potassium chlorate, KClO_3 , that must decompose to produce potassium chloride, KCl , and 1.0 moles of oxygen gas.
- In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 0.40 g of magnesium?
- Ethanol, $\text{C}_2\text{H}_5\text{OH}$, reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.00 moles of ethanol?
- Determine the mass of sodium chloride produced when 0.70 g of sodium (Na) reacts with sodium hydroxide according to the following equation:
 $\text{NaClO}_3 + 2 \text{NaOH} \rightarrow \text{NaClO}_2 + 2 \text{NaOH}$
- In the copper-silver nitrate lab copper metal and silver nitrate solution reacted to produce silver metal and copper(II) nitrate in solution.
 A student placed a copper wire with a mass of 0.05 g in the reaction test tube. The silver-silver nitrate solution contained 0.40 g of silver nitrate.
 She obtained 0.07 g of silver metal. Calculate the percent yield of silver.
- When hydrochloric acid (HCl) is added to sodium hydroxide solution, the products are water, aqueous sodium chloride and sodium chloride gas. What is the per cent yield of 0.04 g of Cl_2 gas collected when 0.01 g of sodium hydroxide solution reacts with excess HCl?
- Phosphorus and bromine react (phosphoryl bromide) to form phosphorus tribromide. If 0.01 g of phosphorus and 0.01 g of bromine react, how many grams of PBr_3 could be produced?
- Iron sulfate and oxygen gas react to form iron oxide and sulfur dioxide. Determine the amount of SO_2 that should be produced in a reaction between 0.01 g of FeSO_4 and 0.01 g of oxygen. What is the percent of the experiment?

$$1. \quad 1.2 \text{ moles of } \text{KClO}_3 \quad 2. \quad 0.001 \text{ moles of } \text{HCl} \quad 3. \quad 0.001 \text{ g } \text{H}_2\text{O} \quad 4. \quad 0.001 \text{ g of } \text{NaClO}_2$$

$$5. \quad 0.01 \text{ g} \quad 6. \quad 0.001 \text{ g} \quad 7. \quad 0.01 \text{ g of } \text{PBr}_3 \quad 8. \quad 0.001 \text{ g of } \text{SO}_2 \quad 9. \quad 0.001 \text{ g of } \text{FeSO}_4$$

Modeling Chemistry Unit 7 Ws 4

Jiyuan Zhang



Modeling Chemistry Unit 7 Ws 4:

Informatics in Control, Automation and Robotics Honghua Tan, 2012-02-01 Session 1 includes 109 papers selected from 2011 3rd International Asia Conference on Informatics in Control Automation and Robotics CAR 2011 held on December 24 25 2011 Shenzhen China This session will act as an international forum for researchers and practitioners interested in the advances in and applications of Intelligent Control Systems It is an opportunity to present and observe the latest research results and ideas in these areas Intelligent control is a rapidly developing complex and challenging field of increasing practical importance and still greater potential Its applications have a solid core in robotics and mechatronics but branch out into areas as diverse as process control automotive industry medical equipment renewable energy and air conditioning So this session will aim to strengthen relationships between industry research laboratories and universities All papers published in session 1 will be peer evaluated by at least two conference reviewers Acceptance will be based primarily on originality and contribution

Estimates of Residence Time and Related Variations in Quality of Groundwater Beneath Submarine Base Bangor and Vicinity, Kitsap County, W, **Biological Chemistry** Thomas Michael Foley, 1991 *Proceedings: Energy & resource. Process modeling. Process simulation. Process dynamics & control. Computer applications*, 1983 Encyclopedia of Environmental Science and Engineering, Volumes One and Two James R. Pfafflin, Edward N. Ziegler, 2006-01-13 Completely revised and updated Encyclopedia of Environmental Science and Engineering Fifth Edition spans the entire spectrum of environmental science and engineering Still the most comprehensive authoritative reference available in this field the monumental two volume encyclopedia has expanded to include 87 articles on topics ranging from acid *Models of Biopolymers By Ring-Opening Polymerization* Stanislaw Penczek, 2018-01-18 There are a number of methods used to synthetically prepare biopolymers their models and bioanalogous polymers This work approaches the syntheses of the three major groups of biopolymers existing in nature polypeptides polysaccharides and nucleic and teichoic acids by ring opening polymerization Until now this method has never been reviewed uniformly for these three groups The majority of models prepared by ring opening polymerization can not reach the complexity of the actual biological molecules However a better understanding of these biopolymers will aid in the use of such molecules in several fields of application in research and other high technologies where they mimic functions of related biopolymers in living organisms **International Critical Tables of Numerical Data, Physics, Chemistry and Technology** Edward Wight Washburn, 1926 Vibrational Spectroscopy with Neutrons Philip Charles Harry Mitchell, 2005 Inelastic neutron scattering INS is a spectroscopic technique in which neutrons are used to probe the dynamics of atoms and molecules in solids and liquids This book is the first since the late 1960s to cover the principles and applications of INS as a vibrational spectroscopic technique It provides a hands on account of the use of INS concentrating on how neutron vibrational spectroscopy can be employed to obtain chemical information on

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