

## Chemistry – Unit 6 Objectives Describing Substances

<p>1. Candidates can describe properties and structures of Lewis's structures of gases, liquids and solids.</p> <p>a. Draw structures for:</p> <p>i. gases including water</p> <p>ii. liquid molecules</p> <p>iii. solids like ionic structures or giant molecules</p>	<p>a. gases including</p> <p>b.</p> <p>c. liquid gas</p> <p>gas</p> <p>liquid gas</p> <p>solid</p> <p>gas</p> <p>gas</p> <p>gas</p>																																								
<p>2. Identify physical states of compounds</p> <p>a. List 5 substances that are solid</p> <p>b.</p> <p>c.</p> <p>d.</p>	<p>a.</p> <p>b.</p> <p>c.</p> <p>d.</p>																																								
<p>3. Candidates describe physical and chemical properties and physical and chemical changes</p> <p>a. Clearly state all the following as physical or chemical changes</p> <p>b. Clearly state all the following as a physical or chemical property</p>	<table> <tr> <td>a. melting point of water</td> <td>0</td> <td>100</td> <td>0</td> </tr> <tr> <td>boiling point</td> <td>100</td> <td>100</td> <td>100</td> </tr> <tr> <td>melting point</td> <td>100</td> <td>100</td> <td>100</td> </tr> <tr> <td>boiling point</td> <td>100</td> <td>100</td> <td>100</td> </tr> <tr> <td>a. water</td> <td>0</td> <td>100</td> <td></td> </tr> <tr> <td>b. density</td> <td>100</td> <td>100</td> <td></td> </tr> <tr> <td>c. boiling</td> <td>100</td> <td>100</td> <td>100</td> </tr> <tr> <td>boiling point</td> <td>100</td> <td>100</td> <td>100</td> </tr> <tr> <td>melting point</td> <td>100</td> <td>100</td> <td>100</td> </tr> <tr> <td>conductivity</td> <td>100</td> <td>100</td> <td>100</td> </tr> </table>	a. melting point of water	0	100	0	boiling point	100	100	100	melting point	100	100	100	boiling point	100	100	100	a. water	0	100		b. density	100	100		c. boiling	100	100	100	boiling point	100	100	100	melting point	100	100	100	conductivity	100	100	100
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boiling point	100	100	100																																						
melting point	100	100	100																																						
conductivity	100	100	100																																						
<p>4. Candidates can state observation and measurement of the rate of reaction and the rate of reaction in the reaction of calcium carbonate with hydrochloric acid</p> <p>a. Calculate rate of reaction</p> <p>b. Explain how the rate of reaction</p>	<p>a.</p> <p>b.</p>																																								

# Modeling Chemistry Unit 4 Describing Substances

## Objectives Answers

**F Rizvi**



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