

Unit 7 Worksheet 1: Molar-mass calculations

Four marks will be given for each question.

1. Write down balanced chemical equations

a) for ethane to react to give water, carbon and carbon dioxide (ethane is C_2H_6)

b) for methanol to react with oxygen to form carbon dioxide and water.

2. Please fill up the spaces in 3.

3. Hydrogen sulfide gas, which smells like rotten eggs, burns to give two products (water and carbon). Three moles of hydrogen sulfide gas reacted to produce 18 moles of water. How many moles of carbon were formed? How many moles of hydrogen sulfide?

Equation: _____

Water:

Carbon:

Answer:

4. Propane, C_3H_8 , burns to give two products (water and carbon). If 10 moles of propane (molar mass 44 g/mol) burn, how many moles of propane were burned?

Equation:

Water:

Carbon:

Answer:

5. Acetylene, C_2H_2 , has the ability to provide for welding. Acetylene and air (oxygen) react to form carbon dioxide and water. If 10 moles of acetylene react, how many moles of carbon dioxide were formed? How many moles of oxygen gas?

Equation:

Water:

Carbon:

Answer:

Modeling Chemistry U8 Test

M Lipman



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Modeling Chemistry U8 Test Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Modeling Chemistry U8 Test**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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