

MCAT ORIGINAL PAPER

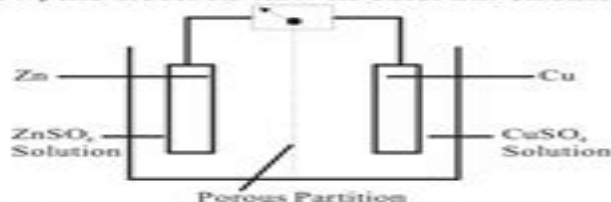
Total MCQs: 220

Max. Marks: 1100

ENTRANCE TEST-2013
For F.Sc. and Non-F.Sc. Students
Time Allowed: 150 Minutes

CHEMISTRY

Q. 45 In the figure given below, the electron flow in external circuit is form:



- A) Zinc to copper electrode
B) Copper to zinc electrode
C) Right to left
D) porous partition to zinc electrode
- Q. 46 Which one of the following is a redox reaction?
A) $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$
B) $2\text{Cl}^- \rightarrow \text{Cl}_2 + 2\text{e}^-$
C) $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$
D) $\text{Na}^+ + 1\text{e}^- \rightarrow \text{Na}$
- Q. 47 The chemical substance, when dissolved in water, gives " H^+ " is called:
A) Neutral
B) Acid
C) Base
D) Amphoteric
- Q. 48 The " pH " of our blood is:
A) 6.7 - 8
B) 7.9
C) 7.5
D) 7.35 - 7.4
- Q. 49 In zero order reaction, the rate is independent of:
A) concentration of the product
B) Temperature of the reaction
C) concentration of the reactant
D) Surface area of the product
- Q. 50 By considering Arrhenius equation, the graph between, " $\frac{1}{T}$ " and " $\log k$ " gives a curve of the type:



- Q. 51 What is the trend of melting and boiling points of the elements of short periods as we move from left to right in a periodic table?
A) Melting and boiling points decrease gradually

Mcat Paper 2013 With Answers

**Melissa J. Margolis, Richard A.
Feinberg**



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