

(iii) Suggest why the mean bond enthalpy of a C=C bond is less than twice the mean bond enthalpy of a C—C bond.

(1)

(iv) Use the mean bond enthalpy data in the table, and the equation given below, to calculate a value for the standard enthalpy change of combustion of propene.

(3)

Bond	Mean bond enthalpy / kJ mol ⁻¹
C=C	612
C—C	347
C—H	413
O=O	498
C=O	805
O—H	464



Answer = _____ kJ mol⁻¹



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Table of Contents January 2015 Accounting Ial Sample Paper

1. Understanding the eBook January 2015 Accounting Ial Sample Paper
 - The Rise of Digital Reading January 2015 Accounting Ial Sample Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying January 2015 Accounting Ial Sample Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an January 2015 Accounting Ial Sample Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from January 2015 Accounting Ial Sample Paper

- Personalized Recommendations
 - January 2015 Accounting Ial Sample Paper User Reviews and Ratings
 - January 2015 Accounting Ial Sample Paper and Bestseller Lists
5. Accessing January 2015 Accounting Ial Sample Paper Free and Paid eBooks
- January 2015 Accounting Ial Sample Paper Public Domain eBooks
 - January 2015 Accounting Ial Sample Paper eBook Subscription Services
 - January 2015 Accounting Ial Sample Paper Budget-Friendly Options
6. Navigating January 2015 Accounting Ial Sample Paper eBook Formats
- ePub, PDF, MOBI, and More
 - January 2015 Accounting Ial Sample Paper Compatibility with Devices
 - January 2015 Accounting Ial Sample Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of January 2015 Accounting Ial Sample Paper
 - Highlighting and Note-Taking January 2015 Accounting Ial Sample Paper
 - Interactive Elements January 2015 Accounting Ial Sample Paper
8. Staying Engaged with January 2015 Accounting Ial Sample Paper
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers January 2015 Accounting Ial Sample Paper
9. Balancing eBooks and Physical Books January 2015 Accounting Ial Sample Paper
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection January 2015 Accounting Ial Sample Paper
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine January 2015 Accounting Ial Sample Paper
- Setting Reading Goals January 2015 Accounting Ial Sample Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of January 2015 Accounting Ial Sample Paper

- Fact-Checking eBook Content of January 2015 Accounting Ial Sample Paper
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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