Question Number	Acceptable Answers	Reject	Mark
23 (d)(i)	2SO₂(g) + 2H₂O(I) + O₂(g) → 2H₂SO₄(aq) ALLOW Multiples M1 - Species and balancing (1) M2 - All state symbols correct M2 can only be awarded for the correct state symbols if M1 has already been awarded OR for a 'nearmiss' equation with the species almost correct (1)		2
	1	1	+
Question Number	Acceptable Answers	Reject	Mark
23 (d)(ii)	CaO is basic / is a base / is a metal oxide OR CaO neutralizes (acidic) SO ₂ / H ₂ SO ₄ OR CaO reacts with a non-metal oxide (SO ₂) OR Basic oxides react with acidic gases ALLOW Alkaline for basic/ alkali for base IGNORE References to forming a salt / formation of calcium sulfate References to the large surface area of powder / effect on rate of reaction		1

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