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Journal of Management Education 34(10)p.1101-1117

Keywords: child sexual abuse; disclosure; self-blame

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ଆବଶ୍ୟକୀୟ ସାମଗ୍ରିକତା ଉପରେ ଆଧାର କରି ସମସ୍ତଙ୍କ ସମ୍ମତିପ୍ରାପ୍ତ ଭାବେ ଉପରୋକ୍ତ ନିୟମାବଳୀ ଗ୍ରହଣ କରାଯିବ । ଏ ନିୟମାବଳୀ ଗ୍ରହଣ ପରେ ଉପରୋକ୍ତ ନିୟମାବଳୀ ଗ୍ରହଣ କରାଯିବ ।

आपका प्रतिक्रिया हमें बताएँ।

১. নিচের কোনটি মৃদব পদার্থ—
(ক) 30 mL 0.1M NH_4OH + 30mL 0.2M HCl
(খ) 40 mL 0.2M CH_3COOH + 30mL 0.3M NaOH
(গ) 50 mL 0.1M NaOH + 40mL 0.2M H_2CO_3
(ঘ) 60 mL 0.1M NaOH + 70mL 0.2M HCl
২. অক্সিগেনের ইলেকট্রন বিন্যাস—
(ক) $1s^2, 2s^2, 2p^4$
(খ) $1s^2, 2s^2, 2p^3$
(গ) $1s^2, 2s^2, 2p^2$
(ঘ) $1s^2, 2s^2, 2p^1$
৩. নিম্নের দ্বারা সংকেত করে—
I. অক্সিজেনের বৃদ্ধি মাত্রা
II. ব্যাসার্ধের আকর্ষণ সঠিক নয় করার মাত্রা
III. নিম্নের একটি নিম্নের
- নিচের কোনটি সঠিক—
(ক) I ও II
(খ) II ও III
(গ) I ও III
(ঘ) I, II ও III
৪. 25° সে. তাপমাত্রায় 150 গ্রাম সম্পৃক্ত দ্রবণে 50 গ্রাম দ্রব দ্রবীভূত থাকলে ঐ দ্রবের দ্রাব্যতা কত—
(ক) 100
(খ) 75
(গ) 50
(ঘ) 25
৫. $2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{SO}_3(\text{g}) + 44.8\text{KCal}$
বিক্রিয়াজনিত তাপমাত্রা বৃদ্ধি করলে—
I. SO_2 এর পরিমাণ হ্রাস পায়
II. K_c এর মান হ্রাস পায়
III. বিক্রিয়া পটভাস্তরী হয়
- নিচের কোনটি সঠিক—
(ক) I ও II
(খ) II ও III
(গ) I ও III
(ঘ) I, II ও III

১৩. কোন যৌগের ক্ষেত্রে $n = 4$, $l = 3$ হলে
- ১৩। কোয়ান্টাম সংখ্যার মান $n = 4$, $l = 3$ হলে
অবস্থানটি হবে—
(ক) $4s$
(খ) $4p$
(গ) $4d$
(ঘ) $4f$
- ১৪। যিহের কোন উপস্থাপন দ্বারা খালি সংরক্ষণ পরীক্ষিত
কিউবিং হলে?
(ক) স্ফটিকের তৈল
(খ) তরল
(গ) কঠিন
(ঘ) ক্রিস্টাল
- ১৫। পৃষ্ঠী অক্সিজেন প্রদানের pH পরিসরে 3-5 বা 6-8 হলে কোন
প্রকারটি দ্বিতীয় প্রকার অণুকে কত গুলি বেশি অক্সিজেন?
(ক) 20 গুলি
(খ) 50 গুলি
(গ) 100 গুলি
(ঘ) 1000 গুলি
- ১৬। M^{3+} আয়নে 23টি ইলেকট্রন থাকলে M এর পারমাণবিক
সংখ্যা কত?
(ক) 23
(খ) 24
(গ) 25
(ঘ) 26
- ১৭। কঠিন বৌদ কোয়ামি
(ক) $TiCl_4$
(খ) MnO_2
(গ) $ScCl_3$
(ঘ) $ZnSO_4$
- ১৮। অক্সিজেনের সঙ্কটের পানির পানির প্রকার কোয়ামি?
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