

ENTRANCE TEST – 2011**CHEMISTRY**

Q.45 In mass spectrometer, detector or collector measures the:

- A) Masses of isotopes
- B) Percentages of isotopes
- C) Relative abundances of isotopes
- D) Mass numbers of isotopes

Q.46 How many 'Cl' (chlorine) atoms are in two moles of chlorine?

- A) $2 \times 6.02 \times 10^{23}$ atoms
- B) $35.5 \times 6.02 \times 10^{23}$ atoms
- C) 2×10^{23} atoms
- D) $2 \times 6.02 \times 10^{23}$ atoms

Q.47 Melting point of water is higher than petrol, because intermolecular forces in water are:

- A) Weaker than petrol
- B) Stronger than petrol
- C) Same as in petrol
- D) Negligible

Q.48 DNA molecule is double stranded, in which two chains of DNA are twisted around each other by:

- A) Hydrogen bonds
- B) Vander Waal's force
- C) Covalent bonds
- D) Dative bonds

Q.49 The elements for which the value of ionization energy is low, can:

- A) Gain electrons readily
- B) Gains electron with difficulty
- C) Loss electrons less readily
- D) Lose electrons readily

Q.50 The nature of cathode rays in discharge tube:

- A) Depends on the nature of gas taken in the discharge tube
- B) Depends upon the nature of cathode in discharge tube
- C) Is independent of the nature of the gas in discharge tube
- D) Depends upon the nature of anode in the discharge tube

Q.51 The ability of an atom in a covalent bond to attract the bonding electrons is called:

- A) Ionization energy
- B) Ionic bond energy
- C) Electronegativity
- D) Electron affinity

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