Digities

Chemical Quantities



THE MOLE AND QUANTIFYING MATTER

10.1 The Mole: A Measurement of Matter

For students using the Foundation edition, assign problems 1–15.

Essential Understanding The mole represents a large number of very small particles.



Frayer Model The Frayer Model is a vocabulary development tool. The center of the diagram shows the concept being defined, while the quadrants around the concept are used for providing the details. Use this model when you want to understand a vocabulary term in more detail.

As you read Lesson 10.1, use the Frayer Model below. Complete the diagram for the term mole.

Definition in your own words	Facts/characteristics
A mole is 6.02 × 10 ²¹ of anything.	A mole contains Avogadro's number of anything. The mass of a mole of an element is the atomic mass of the element in grams.
How moles are used Conversion factors can be	Examples
used to change the number of particles to the number of moles or	1 mol Na = 23.0 g Na = 6.02 × 10 ²³ Na atoms.
the number of moles to the number of particles.	1 mol $H_2O = 18.0 \text{ g } H_2O$ = $6.02 \times 10^{43} H_2O$ molecules.

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Whispering the Secrets of Language: An Mental Quest through Lesson 6 Reading And Study Workbook Copyright

In a digitally-driven earth where displays reign great and instant communication drowns out the subtleties of language, the profound secrets and emotional nuances hidden within phrases frequently move unheard. However, nestled within the pages of **Lesson 6 Reading And Study Workbook Copyright** a charming literary treasure pulsing with fresh thoughts, lies an exceptional quest waiting to be undertaken. Penned by a talented wordsmith, that wonderful opus invites viewers on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within ab muscles cloth of every word. Within the psychological depths of the moving evaluation, we can embark upon a sincere exploration of the book is core subjects, dissect their interesting publishing fashion, and succumb to the strong resonance it evokes deep within the recesses of readers hearts.

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