# Finding the empirical formula

Example: What is the empirical formula of H<sub>2</sub>O<sub>2</sub>?

Answer: Divide everything by the smallest number in the formula – in this case 2. Dividing through by 2 gives HO.

## TASK: Find the empirical formulas of the following:

- 1. C<sub>6</sub>H<sub>6</sub> CH
- C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> CH<sub>2</sub>O
- 3. C<sub>4</sub>H<sub>8</sub> CH<sub>2</sub>
- 4. H<sub>2</sub>O H<sub>2</sub>O
- 5. H<sub>4</sub>C<sub>4</sub>O<sub>8</sub> H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>
- 6. P<sub>4</sub>O<sub>10</sub> P<sub>2</sub>O<sub>5</sub>

- 7. Ca(OH)<sub>2</sub> CaO<sub>2</sub>H<sub>2</sub>
- 8. (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> N<sub>2</sub>H<sub>8</sub>CO<sub>3</sub>
- 9.  $Mg(NO_3)_2$   $MgN_2O_6$

### **How To Calculate Emperical And Molecular Fomula S**

**Christopher O. Oriakhi** 

#### **How To Calculate Emperical And Molecular Fomula S:**

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Concepts And Problems In Physical Chemistry P.S. Raghavan,1997 Contents Introduction Atoms Molecules and Formulas Chemical Equations and Stoichiometry Aqueous Reactions and Solution Stoichiometry Gases Intermolecular Forces Liquids and Solids Atoms Structure and the Periodic Table Chemical Bonding Chemical Thermodynamics Solutions Chemical Kinetics Chemical Equilibrium Acids and Bases Ionic Equilibria I Ionic Equilibria II Redox Reactions Electrochemistry Nuclear Chemistry — Chemistry in Quantitative Language Christopher O. Oriakhi,2021-09-24 Problem solving is one of the most challenging aspects students encounter in general chemistry courses leading to frustration and failure Consequently many students become less motivated to take additional chemistry courses after the first year This book tackles this issue head on and provides innovative intuitive and systematic strategies to tackle any type of calculations encountered in chemistry The material begins with the basic theories equations and concepts of the underlying chemistry followed by worked examples with carefully explained step by step solutions to showcase the ways in which the problems can be presented The second edition contains additional problems at the end of each chapter with varying degrees of difficulty and many of the original examples have been revised

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