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web log H₁₀ 4.05891 × 10⁻⁵ mol dm⁻³ log pK_a pH 3.7540503010030050 and HCOOH 0.1 so HCOO⁻ 0.200 mol dm⁻³ amount vol x

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web only the better candidates could write successive ionisation equations in part c while part d about the pH of equimolar strong and weak acid solutions was poorly answered

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web section 2 pH of strong acids number of protons released monoprotic acid acid that releases one H⁺ ion per molecule e.g.

hcl hydrochloric acid hno₃ nitric acid ch

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