

5. For the equilibrium,  $A(g) \rightleftharpoons B(g)$ ,  $\Delta H$  is  $-40 \text{ kJ/mol}$ . If the ratio of the activation energies of the forward ( $E_f$ ) and reverse

( $E_b$ ) reactions is  $\frac{2}{3}$  then :

- (1)  $E_f = 60 \text{ kJ/mol}$  ;  $E_b = 100 \text{ kJ/mol}$
- (2)  $E_f = 30 \text{ kJ/mol}$  ;  $E_b = 70 \text{ kJ/mol}$
- (3)  $E_f = 80 \text{ kJ/mol}$  ;  $E_b = 120 \text{ kJ/mol}$
- (4)  $E_f = 70 \text{ kJ/mol}$  ;  $E_b = 30 \text{ kJ/mol}$

6. Determination of the molar mass of acetic acid in benzene using freezing point depression is affected by :

- (1) dissociation
- (2) association
- (3) partial ionization
- (4) complex formation

7. The increase of pressure on ice = water system at constant temperature will lead to :

- (1) no effect on the equilibrium
- (2) a decrease in the entropy of the system
- (3) a shift of the equilibrium in the forward direction
- (4) an increase in the Gibbs energy of the system

8. At 298 K, the standard reduction potentials are 1.51 V for  $\text{MnO}_4^-/\text{Mn}^{2+}$ , 1.36 V for  $\text{Cl}_2/\text{Cl}^-$ , 1.07 V for  $\text{Br}_2/\text{Br}^-$ , and 0.54 V for  $\text{I}_2/\text{I}^-$ . At pH = 3, permanganate is expected to oxidize :

$$\frac{RT}{F} = 0.059 \text{ V}$$

- (1)  $\text{Cl}^-$ ,  $\text{Br}^-$  and  $\text{I}^-$
- (2)  $\text{Cl}^-$  and  $\text{Br}^-$
- (3)  $\text{Br}^-$  and  $\text{I}^-$
- (4)  $\text{I}^-$  only

5. इस साम्यावस्था के लिए,  $A(g) \rightleftharpoons B(g)$ ,  $\Delta H$   $-40 \text{ kJ/mol}$  है। अगर अग्र ( $E_f$ ) और पश्चगामी ( $E_b$ ) अभिक्रियाओं के सक्रियण ऊर्जाओं का अनुपात

$\frac{2}{3}$  है, तब :

- (1)  $E_f = 60 \text{ kJ/mol}$  ;  $E_b = 100 \text{ kJ/mol}$
- (2)  $E_f = 30 \text{ kJ/mol}$  ;  $E_b = 70 \text{ kJ/mol}$
- (3)  $E_f = 80 \text{ kJ/mol}$  ;  $E_b = 120 \text{ kJ/mol}$
- (4)  $E_f = 70 \text{ kJ/mol}$  ;  $E_b = 30 \text{ kJ/mol}$

6. ऐसिटिक अम्ल का मोलर द्रव्यमान बेंजीन में हिमांक अवनमन से निकाले जाने पर जिससे प्रभावित होता है वह है :

- (1) वियोजन
- (2) संगुणन
- (3) आंशिक आयनीकरण
- (4) संकुल बिरचन

7. स्थिर ताप पर, अधिक दाब डालने पर बर्फ = जल समुदाय में :

- (1) साम्यावस्था पर कोई प्रभाव नहीं होता है।
- (2) निकाय की ऐन्ट्रॉपी घटती है।
- (3) साम्यावस्था अग्र दिशा में अग्रसर होती है।
- (4) निकाय की गिब्स ऊर्जा बढ़ती है।

8. 298 K पर मानक अपचयन विभव,  $\text{MnO}_4^-/\text{Mn}^{2+}$  का 1.51 V,  $\text{Cl}_2/\text{Cl}^-$  का 1.36 V,  $\text{Br}_2/\text{Br}^-$  का 1.07 V और  $\text{I}_2/\text{I}^-$  का 0.54 V है। pH = 3 पर  $\text{MnO}_4^-$  किसका उपचयन करेगा ?

$$\frac{RT}{F} = 0.059 \text{ V}$$

- (1)  $\text{Cl}^-$ ,  $\text{Br}^-$  और  $\text{I}^-$
- (2)  $\text{Cl}^-$  और  $\text{Br}^-$
- (3)  $\text{Br}^-$  और  $\text{I}^-$
- (4)  $\text{I}^-$  मात्र

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