

## Part 1: Basic Chemistry

## Blank Periodic Table

Fill in the parts of this periodic table covered by the documentary.

☒ noble metals -

|          |          |                            |           |           |           |           |           |           |           |            |            |            |            |            |            |            |            |                           |          |         |         |          |          |
|----------|----------|----------------------------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|------------|------------|------------|------------|------------|------------|------------|------------|---------------------------|----------|---------|---------|----------|----------|
| 1<br>H   |          |                            |           |           |           |           |           |           |           |            |            |            |            |            |            |            | 2<br>He    |                           |          |         |         |          |          |
| 3<br>Li  | 4<br>Be  |                            |           |           |           |           |           |           |           |            |            |            |            |            |            |            |            | 5<br>B                    | 6<br>C   | 7<br>N  | 8<br>O  | 9<br>F   | 10<br>Ne |
| 11<br>Na | 12<br>Mg |                            |           |           |           |           |           |           |           |            |            |            |            |            |            |            |            | 13<br>Al                  | 14<br>Si | 15<br>P | 16<br>S | 17<br>Cl | 18<br>Ar |
| 19<br>K  | 20<br>Ca | 21<br>Sc                   | 22<br>Ti  | 23<br>V   | 24<br>Cr  | 25<br>Mn  | 26<br>Fe  | 27<br>Co  | 28<br>Ni  | 29<br>Cu   | 30<br>Zn   | 31<br>Ga   | 32<br>Ge   | 33<br>As   | 34<br>Se   | 35<br>Br   | 36<br>Kr   |                           |          |         |         |          |          |
| 37<br>Rb | 38<br>Sr | 39<br>Y                    | 40<br>Zr  | 41<br>Nb  | 42<br>Mo  | 43<br>Tc  | 44<br>Ru  | 45<br>Rh  | 46<br>Pd  | 47<br>Ag   | 48<br>Cd   | 49<br>In   | 50<br>Sn   | 51<br>Sb   | 52<br>Te   | 53<br>I    | 54<br>Xe   |                           |          |         |         |          |          |
| 55<br>Cs | 56<br>Ba | 57-71<br>Lanthanide series | 72<br>Lu  | 73<br>Hf  | 74<br>Ta  | 75<br>W   | 76<br>Re  | 77<br>Os  | 78<br>Ir  | 79<br>Pt   | 80<br>Au   | 81<br>Hg   | 82<br>Tl   | 83<br>Pb   | 84<br>Bi   | 85<br>Po   | 86<br>At   | 87-111<br>Actinide series |          |         |         |          |          |
| 87<br>Fr | 88<br>Ra | 89-103<br>Actinide series  | 104<br>Lr | 105<br>Rf | 106<br>Db | 107<br>Sg | 108<br>Bh | 109<br>Hs | 110<br>Mt | 111<br>Uun | 112<br>Uuu | 113<br>Uub | 114<br>Uuq | 115<br>Uup | 116<br>Uud | 117<br>Uus | 118<br>Uue | 119-118<br>Unlabeled      |          |         |         |          |          |

\* Lanthanide series

\*\* Actinide series

|          |          |          |          |          |          |          |          |          |          |          |           |           |           |
|----------|----------|----------|----------|----------|----------|----------|----------|----------|----------|----------|-----------|-----------|-----------|
| 57<br>La | 58<br>Ce | 59<br>Pr | 60<br>Nd | 61<br>Pm | 62<br>Sm | 63<br>Eu | 64<br>Gd | 65<br>Tb | 66<br>Dy | 67<br>Ho | 68<br>Er  | 69<br>Tm  | 70<br>Yb  |
| 89<br>Ac | 90<br>Th | 91<br>Pa | 92<br>U  | 93<br>Np | 94<br>Pu | 95<br>Am | 96<br>Cm | 97<br>Bk | 98<br>Cf | 99<br>Es | 100<br>Fm | 101<br>Md | 102<br>No |

\*  
Mendeleev's  
periodic  
table

\*  
Mendel never  
lived  
to see  
these  
found.

Scandium  
gallium  
germanium

Gold - Au

1. Write the number of subatomic particles in gold:
- a. Protons: 79
  - b. Neutrons: 118
  - c. Electrons: 79

2. Give one property of gold.

high density, malleable, electrical conductivity

3. How much gold is in one ton of the mined rock?

102

# Hunting The Elements Answers

**Didier Musso**



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