

## QUESTION 2

|        | Test                                                    | Observation                                                                                                         | Inference                         |
|--------|---------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------|-----------------------------------|
| (a)(i) | C + Conc. $H_2SO_4$ + strong heat                       | A copious brown irritating fume which turned moist blue litmus paper to red                                         | Gas is $NO_2$<br>$NO_3^-$ present |
| (b)(i) | C + water + NaOH in drops, in excess                    | white gelatinous precipitate<br>precipitate dissolved                                                               | $Al^{3+}$ or $Zn^{2+}$            |
| (ii)   | C + water + $NH_3$ in drops, in excess                  | white gelatinous precipitate<br>precipitate dissolved                                                               | $Al^{3+}$ or $Zn^{2+}$            |
| (iii)  | C + water + freshly prepared $FeSO_4$ + Conc. $H_2SO_4$ | A brown ring is formed at the junction of the acid and aqueous solution.                                            | $NO_3^-$                          |
| (c)(i) | D + water                                               | D dissolved                                                                                                         | soluble                           |
| (ii)   | Solution D + dil HCl                                    | colourless gas, smells like rotten egg, turns moist blue litmus paper to red and lead (II) ethanoate paper to black | $H_2S$ present                    |
| (iii)  | Solution D + NaOH in drops, in excess                   | white precipitate<br>precipitate insoluble                                                                          | $Ca^{2+}$ or $Pb^{2+}$            |
| (iv)   | solution D + $(NH_4)_2CO_3$                             | white precipitate                                                                                                   | $Ca^{2+}$                         |

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**L Manion**



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