Gay-Lussac's Law

Ex. (3) What would be the new temperature of a rigid container of a gas at 101.3 kPa and 293 K if the gas pressure increased to 115 kPa?

$$\frac{PP_{M}}{T_{1}} = \frac{PP_{2}V_{2}}{T_{2}}$$

$$\frac{(101.3 \text{ kPa}) = (115 \text{ kPa})}{(293 \text{ K})}$$

$$\frac{(X)}{X} = \frac{333 \text{ K}}{X}$$

Gay Lussac S Worksheet Answers

Jessica J Manson

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