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# Gas Laws Worksheet

# Boyle, Charles, Pressure and Combined Gas Laws

#### Boyle's Law Problems:

 $P_1V_1 = P_2V_2$ 1 atm = 760.0 mm Hg = 101.3 kPa= 760.0 torr

- If 22.5 L of nitrogen at 748 mm Hg are compressed to 790 mm Hg at constant temperature. What is the new volume?
- A gas with a volume of 4.0L at a pressure of 205kPa is allowed to expand to a volume of 12.0L.
   What is the pressure in the container if the temperature remains constant?
- What pressure is required to compress 196 0 liters of air at 1 00 atmosphere into a cylinder whose volume is 26 0 liters?
- A 40.0 L tank of ammonia has a pressure of 12.7 kPa. Calculate the volume of the ammonia if its
  pressure is changed to 3.4 kPa while its temperature remains constant.

#### Charles' Law Problems:

 $\frac{V_1}{T_1} = \frac{V_2}{T_2}$ 1 atm = 760.0 mm Hg = 101.3 kPa= 760.0 torr

- 5. A container containing 5.00 L of a gas is collected at 100 K and then allowed to expand to 20.0 L. What must the new temperature be in order to maintain the same pressure (as required by Charles' Law)?
- A gas occupies 900.0 mL at a temperature of 27.0 °C. What is the volume at 132.0 °C?
- If 15.0 liters of neon at 25.0 °C is allowed to expand to 45.0 liters, what must the new temperature be to maintain constant pressure?

#### Pressure (Lussac) Law Problems:

 $P_1/T_1 = P_2/T_3$ 

- 8. Determine the pressure change when a constant volume of gas is heated from 20.0°C to 30.0°C.
- If a gas in a closed container is pressurized from 15 atm to 16 atm and its original temperature was 25°C, determine its final temperature.
- 10. The temperature of a sample of gas in a sealed container at 30.0 kPa is increased from -100°C to 1000°C. Determine the final temperature.

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#### Gas Laws Problems Worksheet J Weston Walch Publisher Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Gas Laws Problems Worksheet J Weston Walch Publisher**, a literary masterpiece that delves deep in to the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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